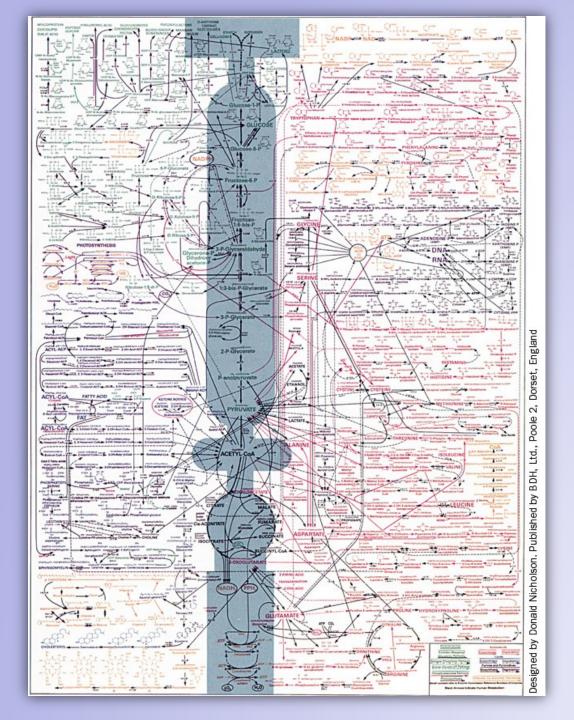
CHEM 539 Molecular Metabolism: Pathways and Regulation Spring 2015

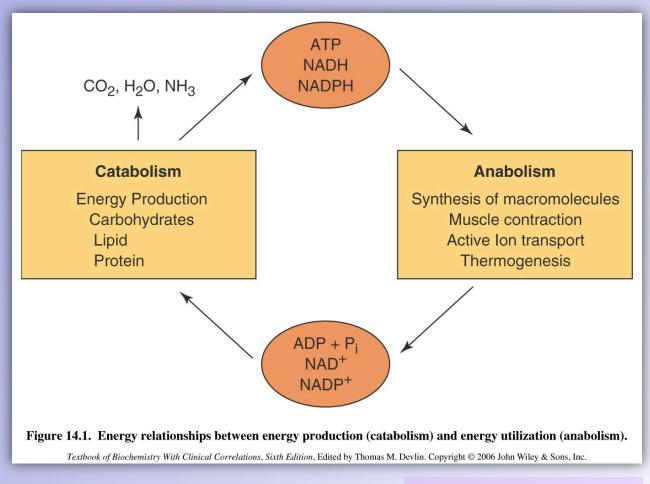
PPT Set 1: Introduction



Map of the major metabolic pathways in a typical cell

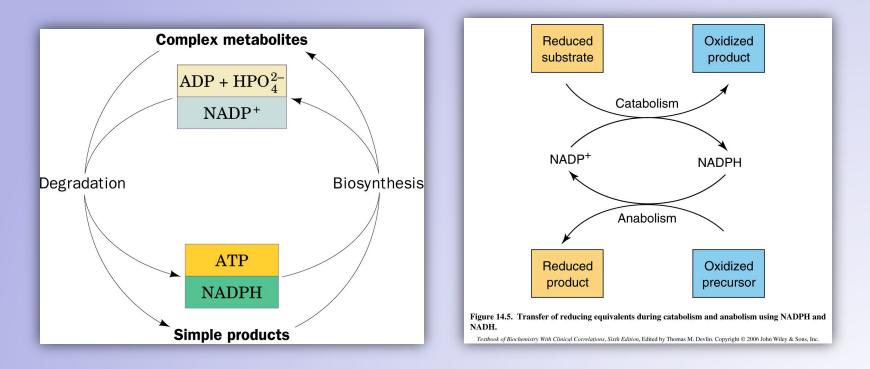
Types of pathways: catabolic anabolic amphibolic cataplerotic anaplerotic

Catabolic pathways are <u>convergent</u>.

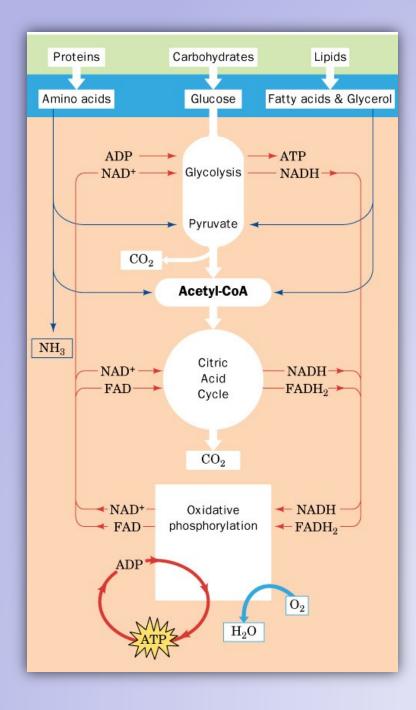


Anabolic pathways are <u>divergent</u>.

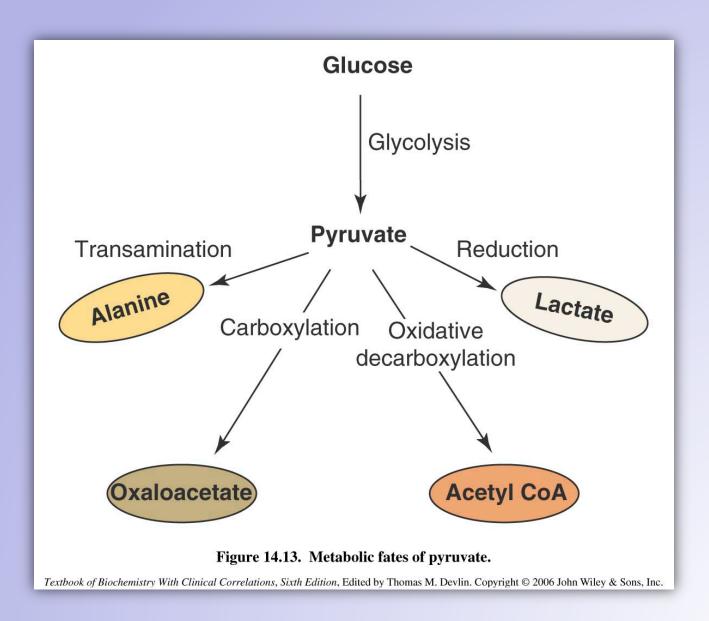
ATP and NADPH are sources of free energy for biosynthetic reactions.

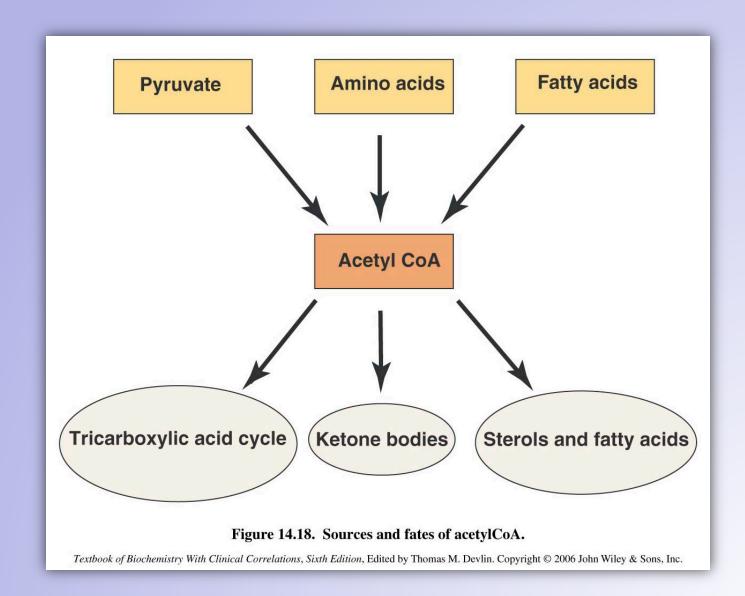


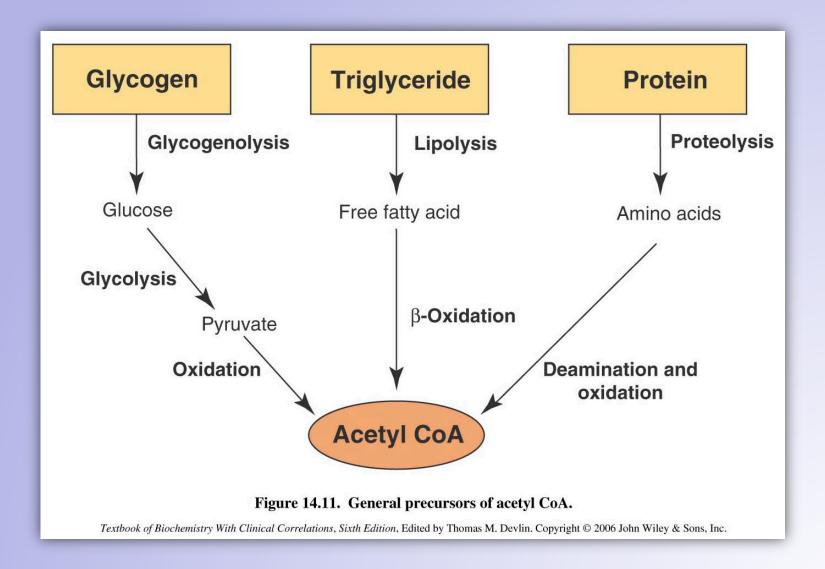
Redox reactions (FAD, NAD⁺) are a major source of free energy in living systems (Nernst equation).



An overview of human catabolism (aerobic)

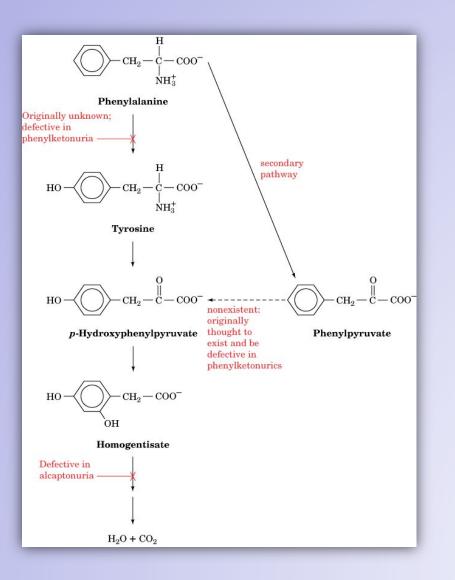






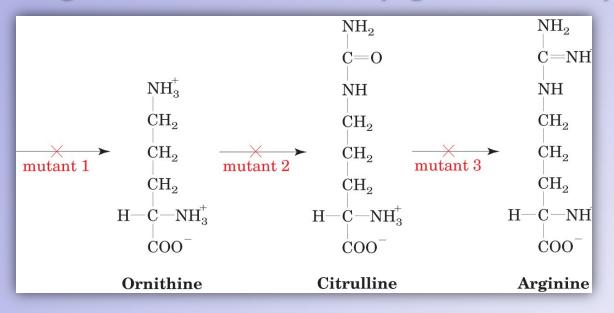
Methodology to elucidate metabolic pathways

Genetic defects cause metabolic intermediates to accumulate (metabolic bottlenecks).



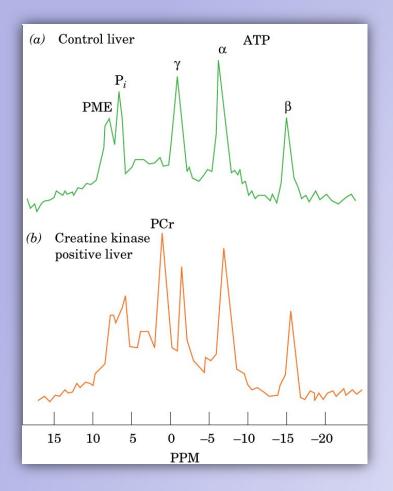
Pathway for phenylalanine degradation in humans

Generating metabolic blocks by genetic manipulation



Pathway of arginine biosynthesis deduced from studies of three arginine-requiring auxotrophic mutants (mutants requiring a specific nutrient for growth). In this case, the mutants were isolated after X-ray irradiation.

All mutants grow in the presence of arginine. Mutant 1 also grows in the presence of ornithine or citrulline, and mutant 2 grows in the presence of citrulline. Mutant 3 does not grow on ornithine or citrulline.



Use of transgenic organisms

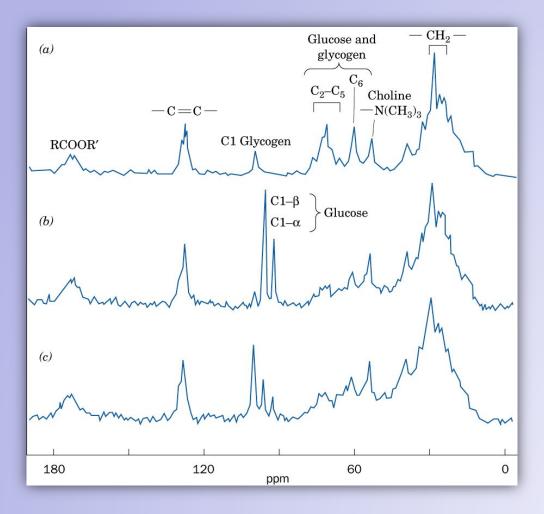
The expression of <u>creatine kinase*</u> in transgenic mouse liver as demonstrated by localized *in vivo* ³¹P NMR.

A. Normal mouse liver after fed a diet supplemented with 2% creatine

B. Mouse liver transgenic for creatine kinase after feeding the same diet.

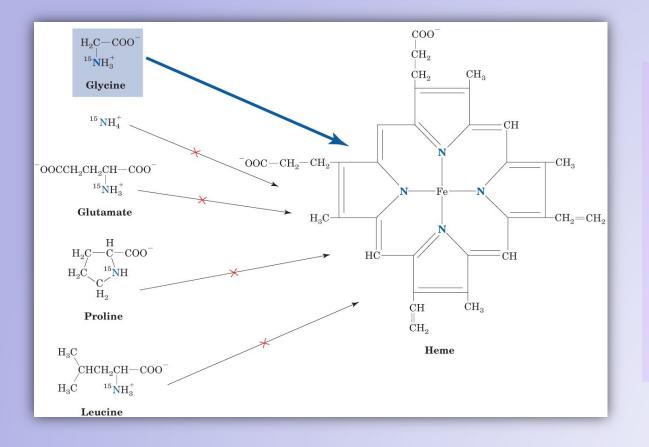
*CK converts creatine into creatine phosphate.

Use of stable isotopes and NMR spectroscopy



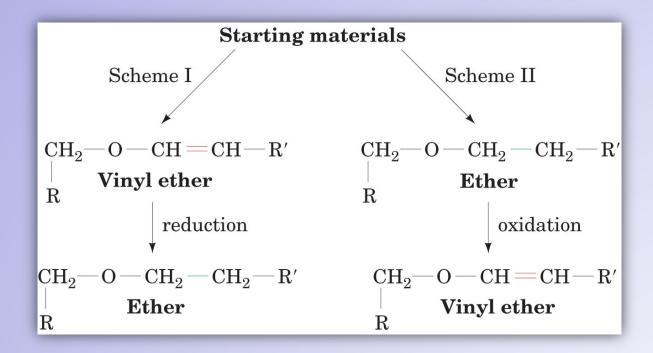
The conversion of D-[1-¹³C]glucose to glycogen as observed by localized *in vivo* ¹³C NMR.

Use of stable isotopes to establish metabolic origins of complex metabolites and precursor-product relationships



The metabolic origin of the nitrogen atoms in heme. Only [¹⁵N] glycine among the various [¹⁵N]labeled precursors tested, serves as a source of nitrogen in the biosynthesis of the porphyrin ring of heme.

Establishing precursor-product relationships (order of appearance of metabolic intermediates)

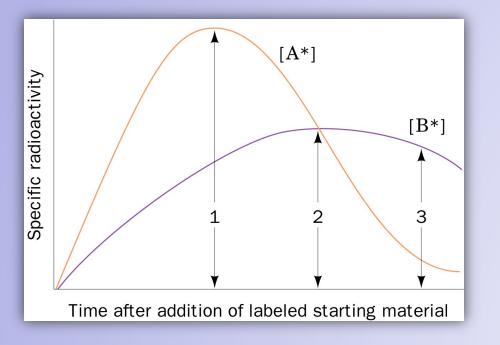


Two possible pathways for the biosynthesis of etherand vinyl ether-containing phospholipids. The pathway shown in Scheme II is the correct pathway. <u>How is this information obtained?</u>

Radioactive tracers: pulse-chase experiments

A pulse of radiolabeled starting material is administered to an organism and the specific radioactivities of the resulting metabolic products are followed over time.

The flow of a pulse of radioactivity from precursor to product.



Criteria that must be met to establish that A* is the precursor of B*:

- 1. While the radioactivity of a product is rising, it should be less than that of its precursor.
- 2. When the radioactivity of a product is at its peak, it should be equal to that of its precursor. The radioactivity of a product peaks <u>after</u> that of its precursor.
- 3. After the radioactivity of a product has peaked, it should remain greater than that of its precursor.

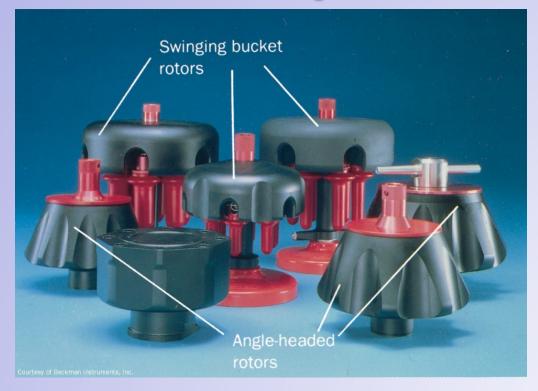
Compartmentation: Eukaryotic cells

Metabolic functions of eukaryotic subcellular organelles

Organelle	Function
Mitochondrion	Citric acid cycle, electron transport and oxidative phosphorylation, fatty acid oxidation, amino acid breakdown
Cytosol	Glycolysis, pentose phosphate pathway, fatty acid biosynthesis, many reactions of gluconeogenesis
Lysosomes	Enzymatic digestion of cell components and ingested matter
Nucleus	DNA replication and transcription, RNA processing
Golgi apparatus	Posttranslational processing of membrane and secretory proteins; formation of plasma membrane and secretory vesicles
Rough endoplasmic reticulum	Synthesis of membrane-bound and secretory proteins
Smooth endoplasmic reticulum	Lipid and steroid biosynthesis
Peroxisomes (glyoxisomes in plants)	Oxidative reactions catalyzed by amino acid oxidases and catalase; glyoxylate cycle reactions in plants

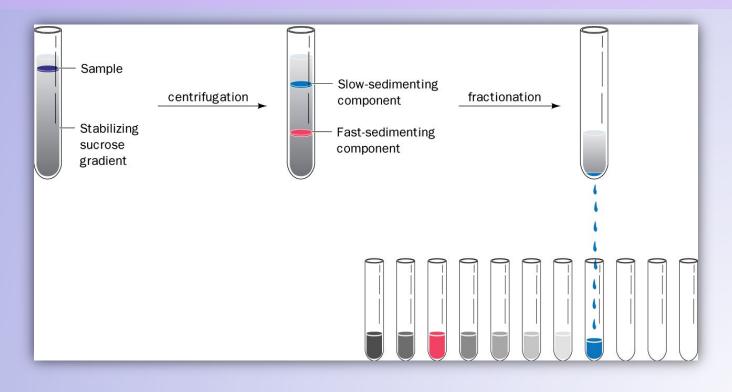
Subcellular organelle isolation in the laboratory

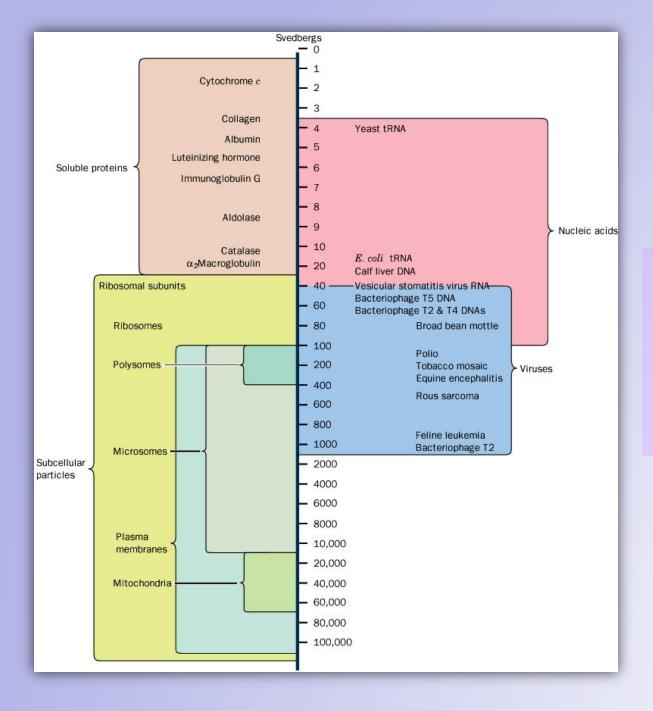
Ultracentrifugation



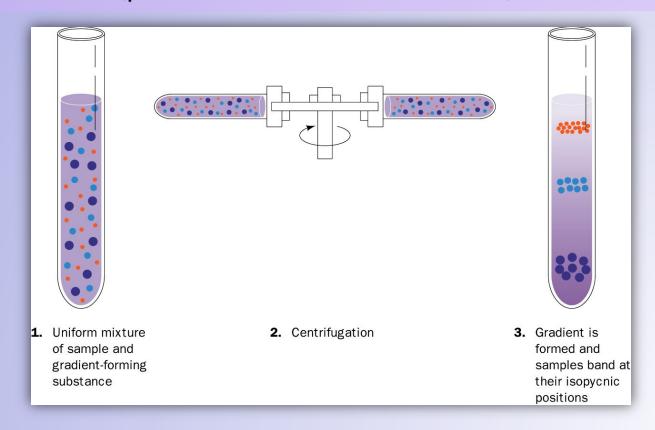
A selection of preparative ultracentrifuge rotors

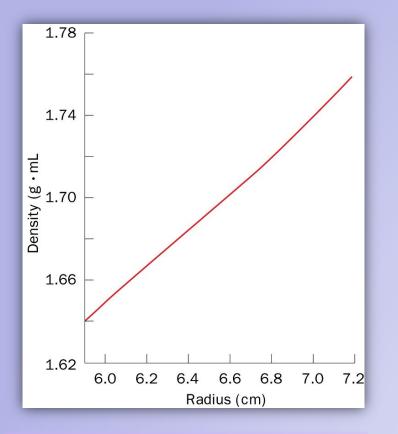
Zonal ultracentrifugation: uses a <u>preformed</u> sucrose density gradient. This method separates similarly shaped macromolecules largely on the basis of their molecular masses (differing sedimentation coefficients).





Sedimentation coefficients in Svedbergs (S) for some biological materials Isopycnic ultracentrifugation (also called equilibrium density gradient ultracentrifugation): Separates particles according to their densities. CsCl or Cs₂SO₄ solutions are spun at high speed to create a density gradient. Sample components band at positions where their densities equal that of the solution. Used for fractionation of subcellular organelles, not for fractionation of protein mixtures (proteins have similar densities).





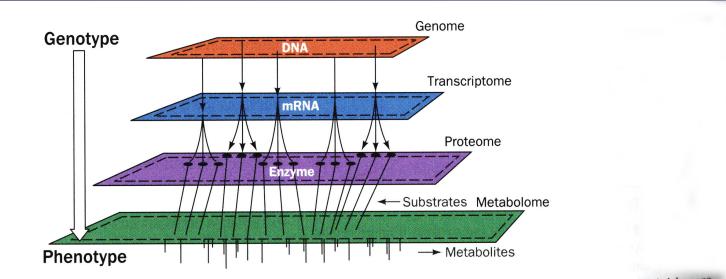
Equilibrium density distribution of a CsCl solution in an ultracentrifuge spinning at 39,460 rpm

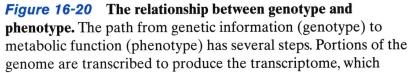
Densities of biological material

Material	Density (g/cm ³)	
Microbial cells	1.05 - 1.15	
Mammalian cells	1.04 - 1.10	
Organelles	1.10 - 1.60	
Proteins	1.30	
DNA	1.70	
RNA	2.00	

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Omics hierarchies in the study of biological metabolism





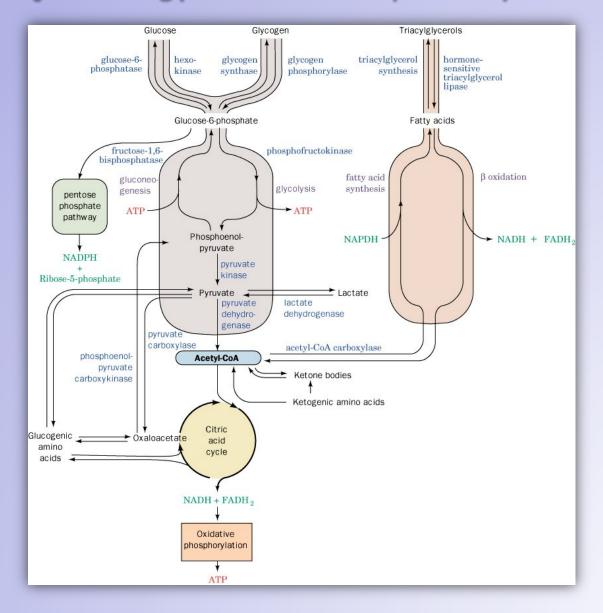
directs the synthesis of the proteome, whose various activities are responsible for synthesizing and degrading the components of the metabolome.

Fuel reserves in a normal 70-kg man

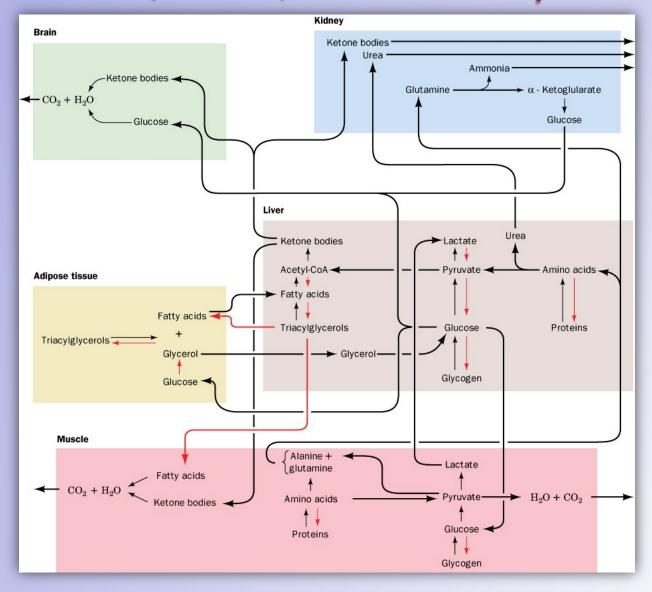
Fuel	Mass (kg)	Calories ^a	
Tissues			
Fat (adipose triacyglycerols)	15	141,000	
Protein (mainly muscle)	6	24,000	
Glycogen (muscle)	0.150	600	
Glycogen (liver)	0.075	300	
Circulating fuels			
Glucose (extracellular fluid)	0.020	80	
Free fatty acids (plasma)	0.0003	3	
Triacylglycerols (plasma)	0.003	30	
Total		166,000	
^{<i>a</i>} One (dieter's) Calorie = 1 kcal = 4.184 kJ.			

Source: Cahill, G.F., Jr., New Engl. J. Med. 282, 669 (1970).

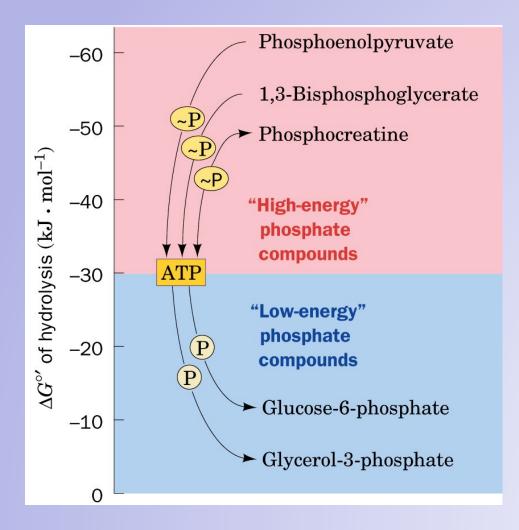
The major energy metabolism pathways in humans



Metabolic interrelationships between brain, adipose tissue, muscle, liver and kidney in humans

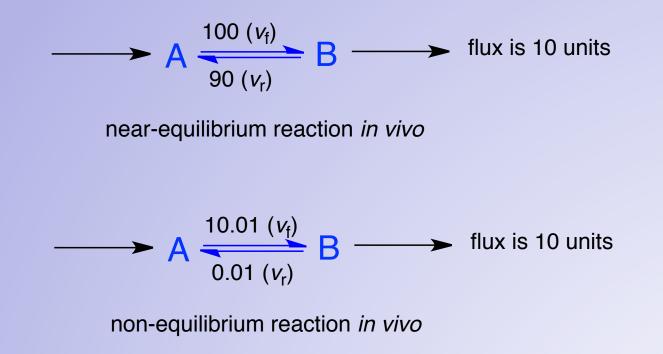


The flow of phosphoryl groups from "high-energy" phosphate donors, via the ATP-ADP system, to "low-energy" phosphate acceptors.



The coupling of catabolism with the generation of high-energy metabolites such as ATP and NADH.

Equilibrium (reversible) and non-equilibrium (irreversible) reactions in metabolic pathways



The net flux through the pathway is given by $(v_f - v_r)$. In the non-equilibrium reaction, the rate of the forward reaction dominates, so that the net flux is almost identical to this rate. In the near-equilibrium reaction, both forward and reverse rates are almost identical but considerably in excess of the flux.