

Beta-enhanced Thermionic Diamond Energy Converters (BTDC) and Nuclear Batteries Project

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Dr. Neil A Fox Project Principal Investigator University of Bristol

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Image : http://www.nasa.gov/mission_pages/sunearth/news/News090811-X1.8flare.html



- PROJECT SCOPE & OBJECTIVES
- SUPPORTING WORK
- PROJECT RESULTS





K Research Programme







K Thermionic Diamond Converter (TDC)

- Solid State Conversion of Heat / Radiation to Electrical Power
- Function of Components Parts
 - Emitter /Cathode
 - Photon absorber (input side)
 - Thermionic emitter (output side)
 - Collector /Anode
 - Electron absorber (input side)
 - Heat sink (output side)







Project objectives

- High current thermionic emission from diamond
- High efficiency radiation absorbers
- Diamond solar energy converter
- Beta-enhanced thermionics
- Beta-diamond nuclear battery







K CVD Diamond

- Material Requirements for Thermionic Diamond
 - Low Work Function surfaces with Stable Negative Electron Affinity
 - High Electrical Conductivity using efficiently doped n-type layers



Emitter Surface - C(100)



Collector surface – C(111)





K Diamond Surface Studies - Modelling

 Oxygen forms stronger bonds with Li than C does. On the other hand, O-terminated surfaces have a very high PEA (high negative surface charge due to lone pairs).



In the optimal structure, the Li-O displacement vectors are almost oriented parallel to the surface, not perpendicular as required for a traditional dipole.



Calculated dipole is large – the NEA is more than twice that of the hydrogenated surface at -4.5 eV.



K.M. O'Donnell et.al., Phys.Rev.B 82, 115303 (2010)

W UV Photoelectron Spectroscopy of SC diamond, B-doped C(100):O,Li



after annealing at 1000°C. Some breakdown is evident above1200°C.





'Photoelectron emission from lithiated diamond', K. M O'Donnell, T.L Martin, M.T Edmonds, A.Tadich, L.Thomsen, J. Ristein, C.I Pakes, N. A Fox, L. Ley, Physica Status Solidi (a), 2014. DOI 10.1002/pssa.201431414

Work function of CVD Diamond films

- NanoESCA shows the average photo-electric work function of boron-doped , ptype diamond is ~4.05eV
- For nitrogen doped, n-type (1.7eV level) ranges from 2.0eV to 2.8eV
- Thermionic emission will be insignificant at temperatures accessible to solar heating (mA/cm² instead of A/cm²)
- How can net flux of emission current be increased?



p-type





Use of energy-filtered photoelectron emission microscopy and Kelvin probe force microscopy to visualise work function changes on diamond thin films terminated with oxygen and lithium monolayers for thermionic energy conversion', *N. A.Fox, et.al., Int. J. Nanotechnol., Vol. 11, Nos. 9/10/11, 2014.*

Weight UHV Scanning Kelvin Probe Microscopy



Work function values of boron doped diamond with Li-O surface termination superimposed onto a surface plot of the same area

 Negative Electron Affinity: vacuum level under conduction band Possible to see work function differences between crystals



Negative Electron Affinity values of boron doped diamond with Li-O surface termination superimposed onto a surface plot of the same area





Modelling Co-doped Diamond







Li-N co-doped diamond films - CASTEP code simulations

Substitutional defect

1:3 Li-to-N ratio

- Formation Energy: 3.88 eV
- Dopant concentration
 - 1.56% for Li (1 atom of Li in 64 atom supercell)
 - 4.69% for N (3 atoms of N in a 64 atom supercell)
- Fermi Level (E_F) 0.5 eV below E_c.
- Dopant level 1.90 eV below E_c.







1:4 Li-to-N ratio

- Formation Energy: 4.88 eV
- Dopant concentration
 - 1.56% for Li (1 atom of Li in 64 atom supercell)
 - 6.25% for N (4 atoms of N in a 64 atom supercell)
- Fermi Level (E_F) <0.1 eV below E_c.
- Dopant level <0.1 eV below E_c.
- The E_F and dopant energy level lying within and just below the conduction band.







Li-N co-doped diamond films - Hot Filament CVD

- $\sim 1.1 \times 10^{19}$ atoms of Li/cm⁻³
- $\sim 2.1 \times 10^{20}$ atoms of N/cm⁻³
- Li:N 1:19
- Properties dominated by N atoms





University of BRISTOL

M. Z.Othman, P.W.May, N.A.Fox, P. J. Heard, Diamond & Related Materials 44 (2014) 1–7.

Li-N (1:19) co-doped diamond films -Thermionic Emission

Co-doped active layer:

- Microcrystalline CVD diamond
- Hydrogen-terminated surface





Measurements taken at ASU 2013

R. J. Nemanich, F. A.M. Koeck. Tianyin Sun,

 $T_{threshold} = 800 \text{ K}$ $T_{max} = 900 \text{ K}$



& Z.Othman

🖌 DTC Test Rigs





Laser-heated Thermionic Set-up

NSQI- Ultra-quiet Laboratory UHV SPM with Radiative heating











Electrically-heated Thermionic Set-up Solar TDC Test Rig





K Laser-heated Temperature Calibration

Two temperature measurement methods:

- Two colour Pyrometer High temp range
- Thermocouple : Low temp range



Heating test of free-standing doped diamond







K Laser-heated TE - Preliminary Results

Single elemental dopant:

- Microcrystalline CVD Diamond (MCD)
- Metal Substrate
- Hydrogen-terminated surface
- N content ~1 x 10²⁰ atoms.cm⁻³





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$$T_{threshold} = 673 \text{ K}$$

• $T_{max} = 963 \text{ K}$





K Laser-heated TE - Preliminary Results

Co-doped active layer:

- Metal substrate
- Hydrogen-terminated surface
- Li:B ratio TBD









K Laser-heated TE - Preliminary Results

Co-doped active layer:

- Free-standing e6 BDD substrate
- Hydrogen-terminated surface
- Li:N ratio 1:19









K Laser-heated TE - Summary







K Isotopic Diamond Converters

- Use of ¹³C & ¹²C diamond
 - Thermal carrier confinement
- Use of ionising ¹⁴C diamond
 - Generates e-h pairs in bulk
 - Potential to generate high SEE at a diamond surface
- Potential outcomes
 - Increased flux of thermionic carriers leaving hot cathode
 - New electrode architectures







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Li-N co-doped diamond films + O,Li termination - Thermionic Emission







NanoESCA Mk II system configuration for Bristol







Ke NanoESCA Laboratory Specification

- Focussed Light sources
 - Energy Resolution in PEEM modes
 - VUV : 50 (28) meV
 - o X-ray : 190 (100) meV
 - Lateral edge resolution in PEEM
 - o VUV: 60 (20) nm
 - X-ray : 480 (320) nm









Li-N co-doped diamond films - CASTEP code simulations

Interstitial defect

1:1 Li-to-N ratio (next to each other)

- Formation Energy 9.41 eV
- Dopant concentration
 - 1.56% for Li (1 atom of Li in 64 atoms supercell)
 - 1.56% for N (1 atoms of N in 64 atoms supercell)



1:1 Li-to-N ratio (far apart)

- Formation Energy 11.32 eV
- Dopant concentration
 - 1.56% for Li (1 atom of Li in 64 atoms supercell)
 - 1.56% for N (1 atoms of N in 64 atoms supercell)



- From the formation energy calculated. N atom act as a trap to pin down Li atom and this reduced the mobility of Li in diamond at high temperature.
- This concept is suitable to immobilise Li in diamond lattice.



