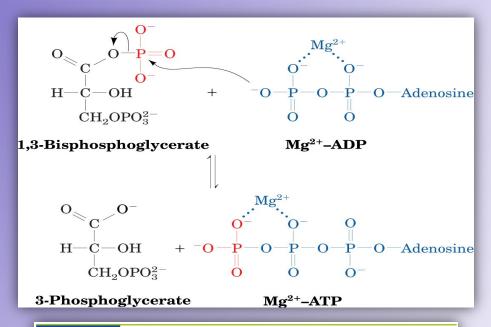
CHEM 539

Molecular Metabolism: Pathways and Regulation Spring 2015

PPT Set 2b

Glucose transport; glycolysis; degradation of other monosaccharides; alternatives to glycolysis

Mechanism of the PGK reaction

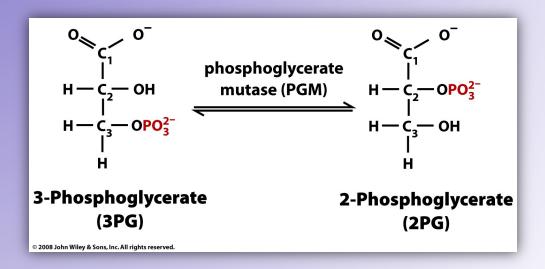


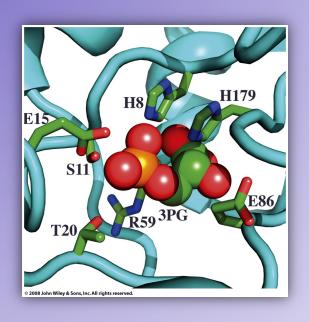
		$\Delta G^{\circ\prime}$	ΔG
Reaction	Enzyme	(kJ⋅mol ⁻¹)	(kJ·mol ⁻¹)
1	Hexokinase	-20.9	-27.2
2	PGI	+2.2	-1.4
3	PFK	-17.2	-25.9
4	Aldolase	+22.8	-5.9
5	TIM	+7.9	~0
6 + 7	GAPDH + PGK	-16.7	-1.1
8	PGM	+4.7	-0.6
9	Enolase	-3.2	-2.4
10	PK	-23.0	-13.9

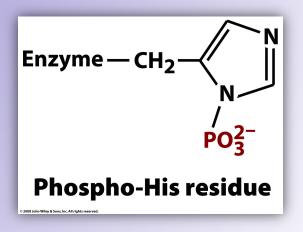
Wiley (1973).

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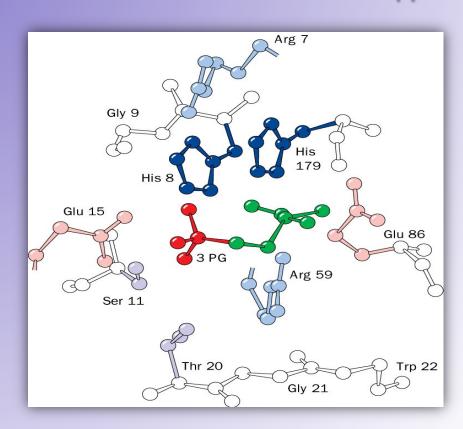
3-Phosphoglycerate mutase



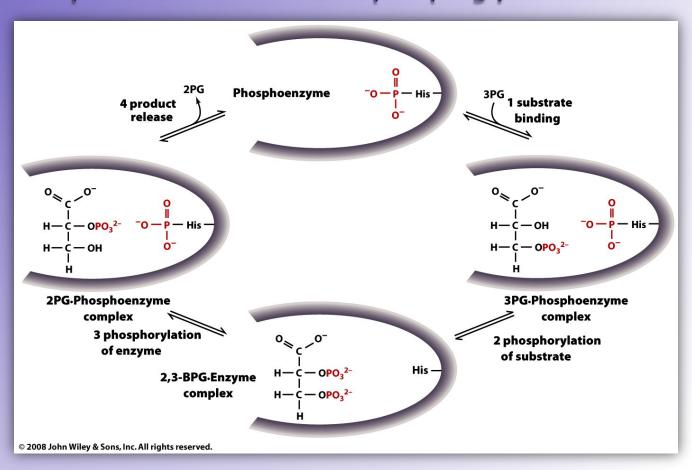




The active site region of yeast phosphoglycerate mutase (dephospho form) showing the substrate, 3-phosphoglycerate, and some of the side chains that approach it

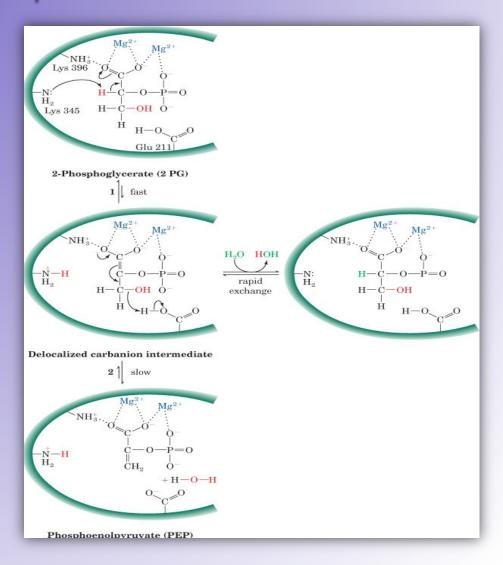


Proposed mechanism of 3-phosphoglycerate mutase



Enolase

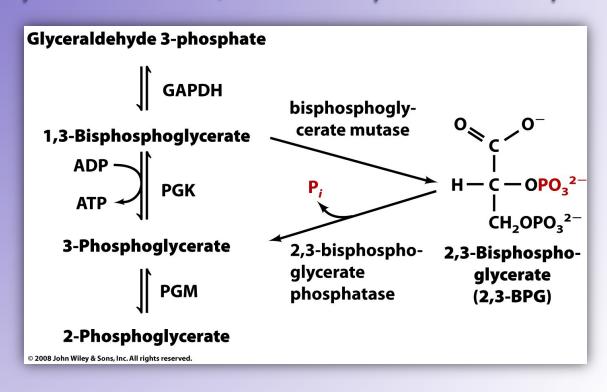
Proposed reaction mechanism of enolase



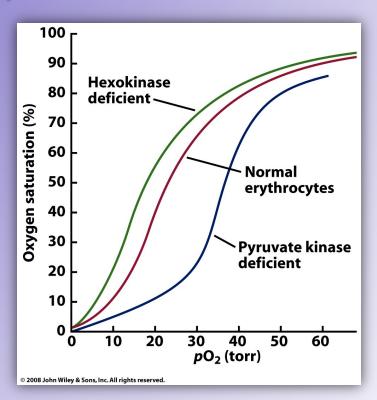
Pyruvate kinase

Second substrate-level phosphorylation reaction of glycolysis

Glycolytic detour: 2,3-BPG biosynthesis in erythrocytes



Impact of 2,3-BPG on O_2 -hemoglobin binding affinity: hexokinase and PK deficiencies



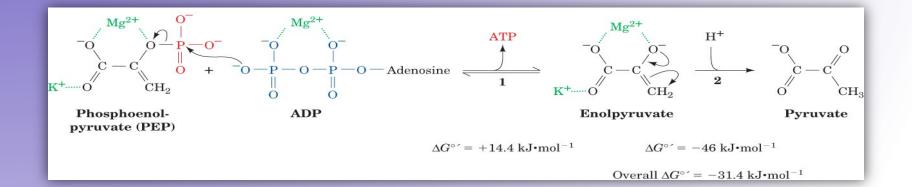
Explanation of the very (-) change in free energy associated with the PK reaction

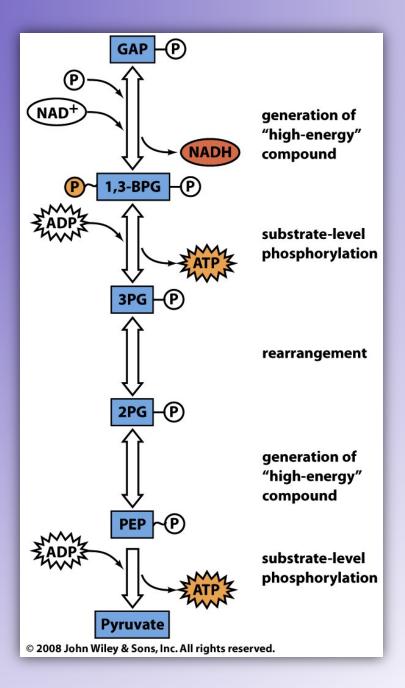
Hydrolysis
$$G^{\circ\prime\prime} = -16 \text{ kJ} \cdot \text{mol}^{-1} \qquad C - O \sim PO_3^2 + H_2O \qquad C - O - H + HPO_4^2$$

$$H \qquad H \qquad H \qquad H \qquad H \qquad H$$
Phosphoenol-pyruvate
$$COO^- \qquad COO^- \qquad COO^- \qquad COO^- \qquad COO^- \qquad H - C - H \qquad H$$

$$G^{\circ\prime\prime} = -46 \text{ kJ} \cdot \text{mol}^{-1} \qquad H \qquad H \qquad H$$
Pyruvate (enol form)
$$G^{\circ\prime\prime} = -61.9 \text{ kJ} \cdot \text{mol}^{-1} \qquad COO^- \qquad COO^- \qquad COO^- \qquad COO^- \qquad HPO_4^2 \qquad H - C - H \qquad H$$

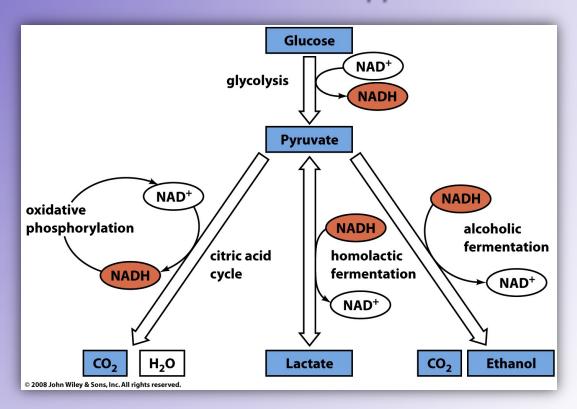
$$O = -61.9 \text{ kJ} \cdot \text{mol}^{-1} \qquad COO^- \qquad COO^- \qquad COO^- \qquad COO^- \qquad HPO_4^2 \qquad H - C - H \qquad H$$





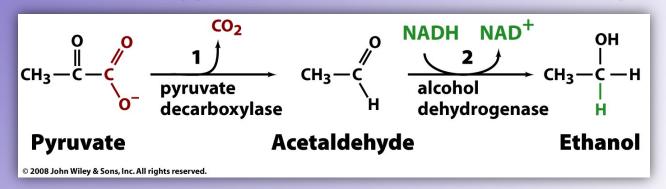
Second half of glycolysis:
ATP-yielding phase

Metabolic fates of pyruvate

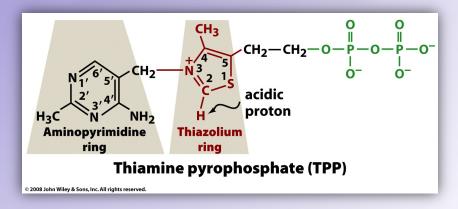


Conversion of pyruvate to lactate: LDH reaction

Conversion of pyruvate to ethanol via acetaldehyde

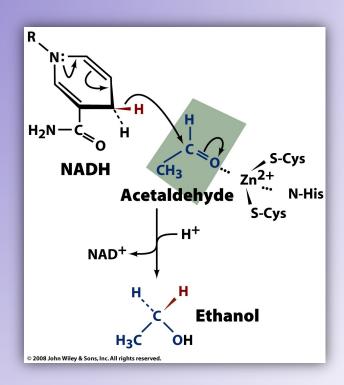


Enzyme-catalyzed decarboxylation of an α -ketoacid (pyruvate): TPP coenzyme is required for charge delocalization



Proposed mechanism of pyruvate decarboxylase

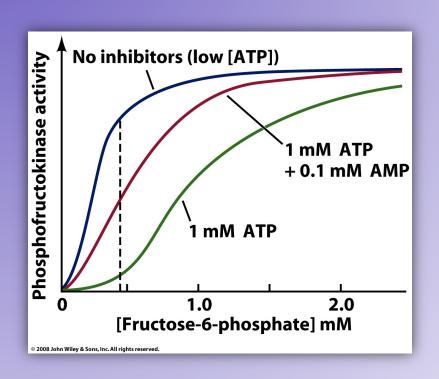
Alcohol dehydrogenase: Stereospecfic transfer of hydride from NADH to acetaldehyde

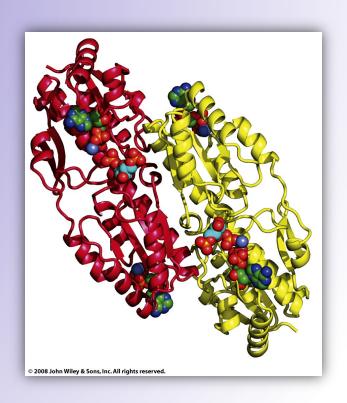


Some effectors of the non-equilibrium enzymes of glycolysis

Enzyme	Inhibitors	Activators ^a
НК	G6P	_
PFK	ATP, citrate, PEP	ADP, AMP, cAMP, FBP,
		F2,6P, F6P, NH_4^+ , P_i
PK (muscle)	ATP	AMP, PEP, FBP
PK (muscle)	ATP	F2,6P, F6P, NH ₄ ⁺ , P _i

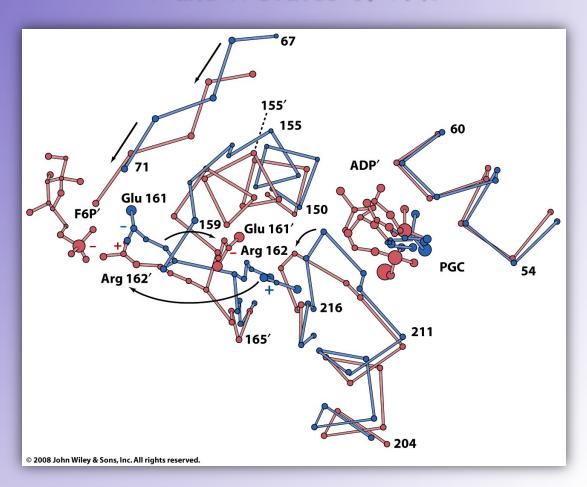
^aThe activators for PFK are better described as deinhibitors of ATP because they reverse the effect of inhibitory concentrations of ATP.



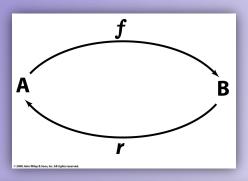


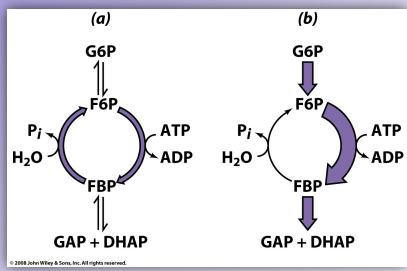
PFK is an allosteric enzyme (tetramer)

T and R states of PFK



Control of glycolytic flux via substrate cycling

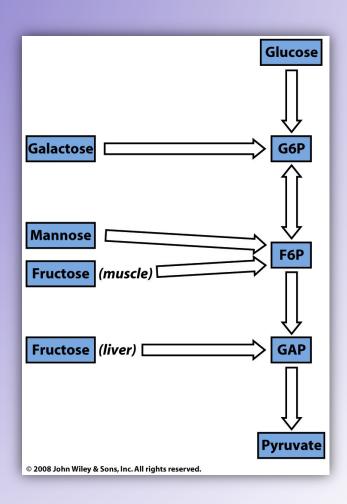




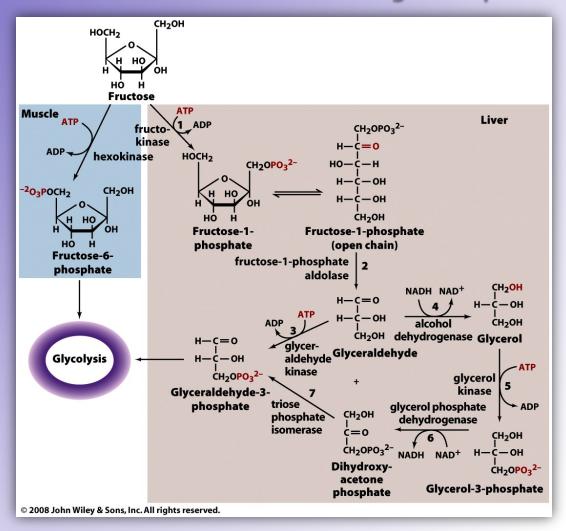
resting muscle

active muscle

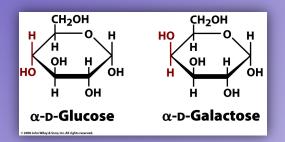
Entry routes of other monosaccharides into glycolysis

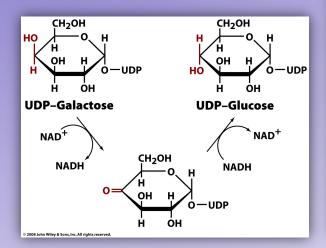


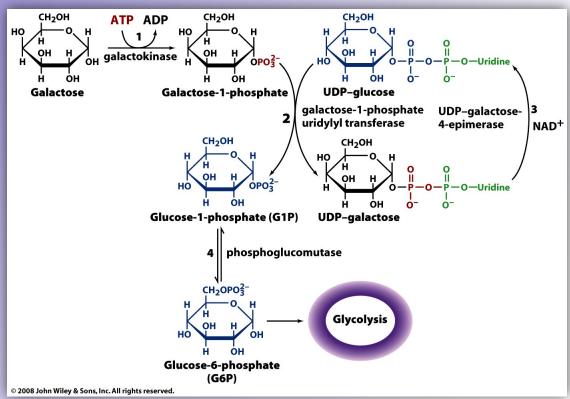
Metabolism of D-fructose is organ-dependent



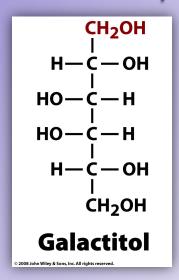
Metabolism of D-galactose (human)



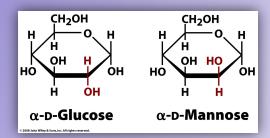


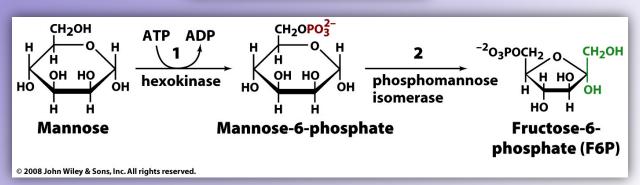


By-product of galactosemia (step 2 deficient) (buildup in lens of the eye - cataracts)



Metabolism of D-mannose (human)





Entner-Doudoroff pathway for glucose breakdown (bacteria)

