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Pyroxenes	Title
James. A. Author and Carol C. Coauthor	Author List
Department of Scientific Studies State University City, State 12345-6789, U.S.A.	Address
j.author@SU.edu; CC.Coauthor@SU.edu	Email addresses
<b>ROCK-FORMING PYROXENES</b>	1st order
Assemblages of natural pyroxenes [no period, no caps]	2nd order
Subsolidius assemblages. (An indented, italicized, bold heading is	the form for a third-

order heading and that the text should follow after it exactly as this text does.) 3rd order

Table 23. The composition      [center heading,	
Constituent	Weight %
Milk	XX
Cocoa	уу
Sugar	ZZ

Do not use too open a format. Maximum dimensions (including headings) are 6" (152 mm) by 9.5" (241 mm). Tables can be in portrait or landscape layout.

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A simple equation may be incorporated into the text, but if it will be referred to again, it is best to type it on a separate line with its number in parentheses, right justified (use tab stop), on the same line. Example:

$$\sum t_1 z = 6.589 [\{T_{1z} - O\} - 1.605]$$
(32)

Use the following style in both the equation <u>and</u> surrounding text: NUMBERS, FUNCTIONS, and CONSTANTS are roman (not italic, not bold) SCALAR VARIABLES are *italicized* VECTORS and MATRICES are **bold faced**  Abbreviations and symbols should be as below. There is a space between all numbers and units:

#### Common abbreviations (note Roman type)

cf. = compare e.g. = for example et al. = and others etc. = and other things i.e. = that is p. = page(s) vs. = versus

## Length (l)

$$\begin{split} & m = meter(s) \\ & cm = centimeter(s) \ (1 \ cm = 10^{-2} \ m) \\ & mm = millimeter(s) \ (1 \ mm = 10^{-3} \ m) \\ & \mu m = micrometer(s) \ (1 \ \mu m = 10^{-6} \ m) \ NOT \ micron \ or \ \mu \\ & nm = nanometer(s) \ (1 \ nm = 10^{-9} \ m) \\ & Å = angstrom(s) \ (1 \ Å = 10^{-10} \ m) \\ & pm = picometer(s) \ (1 \ pm = 10^{-12} \ m) \end{split}$$

#### Mass (m)

g = gram(s) kg = kilogram(s) mg = milligram(s) µg = microgram(s)

#### Volume (V)

L = liter(s)mL = milliliter(s) cm<sup>3</sup> = cubic centimeter(s) Å<sup>3</sup> = cubic angstrom(s)

#### Pressure (P)

Pa = pascal(s) kPa = kilopascal(s) MPa = megapascal(s) GPa = gigapascal(s) bar or bars (no abbreviation) kbar = kilobar(s) atm = atmosphere(s)

#### Temperature (T)

K = Kelvin(s) °C = degrees Celsius [space before degree sign. c.f. angles where there is no space between digit and degree sign]

## **Energy:**

J = joule(s) kJ = kilojoule(s) (preferred) cal = calorie(s) kcal = kilocalorie(s) **Density (D or \rho)** g/cm<sup>3</sup> = gram(s) per cubic centimeter

Concentration (use square brackets [ ] around chemical symbol, e.g., [Ca<sup>-2</sup>]) % = percent [no space prior] % = per thousand [no space prior] wt% = weight percent mol% = mole percent vol% = volume percent M = molar concentration (mol/L) m = molal concentration (mol/kg) ppm = parts per million ppb = parts per billion

#### Time (t)

s = second(s) min = minute(s) h = hour(s) d = day(s) yr = year(s) Ma = million years Ga = billion years (ago or date) m.y. = million years b.y. = billion years (duration)

### Frequency (f or v)

Hz = hertz kHz = kilohertz MHz = megahertz

Miller indices--note that letter symbols are italicized but numeric values are not, e.g., [hkl] but [100] hkl = diffraction symbol (hkl) = face symbol {hkl} = form symbol [hkl] = edge or zone symbol

## **Unit-cell** measurements: *a*, *b*, *c* = edge lengths $\alpha$ , $\beta$ , $\gamma$ = angles **a**, **b**, **c** = vectors--note boldface

## **Optical** measurements: dispersion (r, v), optic axial angle $(2V, 2V_x, 2V_z)$ ; principal vibration directions (E, O; X, Y, Z); refractive indices $(n; \varepsilon, \omega; \alpha, \beta, \gamma)$

## REFERENCES

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### Journal:

Hawthorne FC, Oberti R (2007) Classification of the amphiboles. Rev Mineral Geochem 67:55-88

#### Journal using doi's:

Meier R, Anderson J, Verryn S (2012) Industrial X-ray diffraction analysis of building materials. Rev Mineral Geochem 74:147-165, doi: 10.2138/rmg.2012.74.4

## Article published only on WWW:

Barrick RE, Showers WJ (1999) Thermophysiology and biology of *Giganotosaurus*; comparison with *Tyrannosaurus*. Palaeontologia Electronica, *http://www.palaeo-electronica.org/1999\_2/gigan/issue2\_99.htm* 

#### Book:

Tossell JA, Vaughn DJ (1992) Theoretical Geochemistry: Application of Quantum Mechanics in the Earth and Mineral Sciences. Oxford University Press, New York

### Article in Book:

Keil K (1969) Meteorite composition. In: Handbook of Geochemistry. Vol 1. Wedepohl KH (ed) Springer, Berlin, p 78-115

#### Edited Book:

Schulze DG, Stuicki JW, Bertsch PM (eds) (1999) Synchroton X-ray Methods in Clay Science. The Clay Minerals Society, Boulder Colorado

## Conference Proceedings or Abstract:

Rosso KM, Bylaska EJ, Rustad JR (1999) The energetics of cation exchange in muscovite mica. EOS Trans, Am Geophys Union 80:F1108

### Dissertation/Thesis:

Bartelmehs KL (1993) Modeling the properties of silicates. PhD Dissertation, Virginia Polytechnic Institute & State University, Blacksburg, Virginia

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Smerson XY (2005b) 25 years of research in	Smerson (2005b)
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Smerson XY, Xavier X (1997) Monoclinic ushywushyite in	Smerson and Xavier (1997)
Smerson X, Hibou A, Blye G, Xu H (1992a) Systematics of technetium mineralogy	Smerson et al. (1992a)
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