CE/SC 10110-20110 Mineral Resources





Fifth Edition

Chapter 15



Major Categories of Mineral Resources





Mineral Deposits

Main Subdivisions: Metallic Resources:

• Gold, Silver, Copper, Lead, Zinc, Iron, Aluminium

Non-metallic Resources:

• Sand, Gravel, Gypsum, Halite, Dimension stone

Native Metals

Metals are found in nature: Cu, Au, Ag. Also mixtures/blends of metals or **alloys - Copper + Tin = Bronze**.

Metallic: outer shells of atoms move easily from one to another. Allows electricity to move.





Economic deposits require concentration by geologic processes.

Metallic Deposits

Native metals liberated from rock by heating it: **Smelting**. Metal runs off and leaves behind a non-metallic residue: **Slag**.



Metallic Deposits

Ore: deposit that can be mined economically. Depends upon, political, economic, social factors. The concentration determines the *grade* of the deposit.

Ore is mined that contains native metals or a concentrated accumulation of **ore minerals**. Uneconomic minerals found with native metals/ore minerals is called **Gangue**.



	Common Ore Minerals		
Metal	Ore Mineral	Composition Al ₂ O ₃ •nH ₂ O	
Aluminum	Bauxite (a mineral mixture) re)		
Chromium	Chromite	FeCr ₂ O ₄	
Copper	Native copper	Cu	
	Chalcocite	Cu ₂ S	
	Chalcopyrite	CuFeS ₂	
Gold	Native gold	Au	
Iron	Hematite	Fe ₂ O ₃	
	Magnetite	Fe ₃ O ₄	
Lead	Galena	PbS	
Manganese	Pyrolusite	MnO ₂	
Mercury	Cinnabar	HgS	
Nickel	Pentlandite	(Fe, Ni)S	
Silver	Native silver	Ag	
	Argentite	Ag ₂ S	
Tin	Cassiterite	SnO ₂	
Uranium	Pitchblende	U ₃ O ₈	
	Carnotite	K(UO2)2(VO4)2+3H2O	
Zinc	Sphalerite	ZnS	

What is an Ore?

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Metal	Mineral Name	Chemical Formula
Copper	Chalcocite	CueS
	Chalcopyrite	CuFeS
	Bornite	CityFeSa
	Azurite	Caulcoulz(OH)».
	Malaphite	Cue(CO ₄)(OH) ₂
Tron	Hematite	FegOs
	Magnetite	Fe ₂ O ₄ ;
Tin	Cassiterile	SnQ ₂
Lead	Galena	P6S
Mercury	Cinnabar	Has
Zinc	Sphalerite	ZnS
Aluminum	Kaolinite	Al ₂ Si ₂ O ₅ (OH) ₄
	Corundum	AlgOr
Chrome	Chromite	(Fe Mg)(Cr ALFe)2O ₄
Nickel	Penflandite	(NI,Fe) ₂ S ₈
Titanium.	Rutile	TIO
	Ilmenite	FetiOa
Tungsten	Sheelite	GaWO
Molybdenum	Molybdenite	MoSa
Magnesium	Magnesite	MgCOg
	Dolomite	CaMg(CO ₃) ₈
Manganese	Pyrolusite	MnQ ₂
	Rhodochrosite	MnCO



The difference between an ore and other rock is that metals are concentrated in the ore. 7

How do Ore Deposits Form?

Ore-forming geologic processes:

- Magmatic activity
- Hydrothermal alteration
- Secondary enrichment
- Groundwater transport (MVT)
- · Sedimentary processes
- · Residual weathering
- · Hydraulic sorting



Ore Deposits Magmatic Deposits

Crystal Settling from magma: Cr (chromite), Bushveld, South Africa Sulfide (Sudbury, Ontario) - "Massive Sulfide Deposit".





◆ FIGURE 135 Layered mineral deposit; Red Mountain, Alaska. The black layers are chromite (FeCr₂O₄); the light layers are granitic rocks.

Ore Deposits

Hydrothermal Deposits

Circulation of reactive hot water solutions through a magma or rocks surrounding an igneous intrusion.

Metallic ions are dissolved. They are redeposited when the fluids enter a region of lower temperature / lower pressure / different acidity /different oxygen availability. This creates a hydrothermal deposit.



Hydrothermal Deposits

Disseminated Deposit: ore minerals are dispersed throughout the intrusion/country rock - often called "Massive Sulfide Deposits". Vein Deposit: ore minerals fill joints with in the pre-existing rock.

Hydrothermal copper deposits are often found in porphyritic igneous rocks; these are **Porphyry Copper Deposits**.





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Ore Deposits Hydrothermal Deposits

Black Smokers are forming massive sulfide deposits around midocean ridges present day.



Secondary-Enrichment Deposits



Original deposit is sub-economic. Secondary enrichment processes allow concentration of the ore minerals to economic levels. This is also called **Supergene Enrichment**. These deposits can form when dissolution occurs in oxygenated waters and the water moves to a deoxygenated region.



Ore Deposits

Mississippi-Valley-Type Deposits

Groundwater beneath mountain belts sinks several kilometers and comes back to the surface up to several hundred kilometers away. After sinking, the water heats up and dissolves metals. When it rises into cool rocks, the metals precipitate.

Lead- and zinc-bearing veins are formed in this way - known as **MVT Ores**.



Sedimentary Deposits

Banded Iron Formations (BIFs): Fe^{2+} is soluble in water, but Fe^{3+} is not. These are found in the Proterozoic, contain Hematite (Fe_2O_3) and Magnetite (Fe_3O_4) in layers that alternate with Fe-rich chert (jasper). They present evidence for an oxygenated atmosphere.



Distribution of Mineral Resources

Sedimentary Deposits

Surficial Precipitation:

Evaporite deposits – two types – marine evaporites (salts of Na, K, Ca – halite, sylvite, anhydrite, gypsum and bedded phosphates); non-marine evaporites (Ca & Na carbonates, nitrate, sulfate, and borate minerals).



Sedimentary Deposits

Manganese Nodules: potato-size (25% Manganese, 15% Iron, 2% Nickel, 2% Copper).

Chemistry of seawater in some places promotes the deposition of Mn-oxide nodules.

Mining companies have begun to explore technologies to exploit these deposits as they 720 years worth of Cu and 60,000 years of Mn (at current rates of composition).





Manganese nodules grow slowly on the sea floor and are rich in MnO_2 and trace elements.

Distribution of Mineral Resources

Residual Mineral Deposits

Deep chemical weathering of rock in hot, humid, tropical climates promotes mineral enrichment through leaching – insoluble residue left behind.

Lateritic soils = rich in iron. If iron is low, the mineral **Bauxite** forms, the principle Al mineral.

Secondary enrichment can occur by water moving through, for example, a disseminated sulfide deposit – moves a submarginal grade deposit to an economic one.



◆ FIGURE 13.14 The development of ores by secondary enrichment. Descending ground water oxidizes and dissolves soluble sulfide minerals, carrying them downward and leaving a brightly colored residue of limonite and hematite. Below the water table new metallic minerals are precipitated as 18 secondary ore minerals.



Ore Deposits Sedimentary & residual deposits are found in shield areas of continents. Bauxite forms on granite that has experienced major leaching in tropical areas.



Distribution of Mineral Resources

Ore deposits associated with igneous rocks are associated with plate boundaries or hot spots – tectonic processes are the dominant control. Granitic



Igneous and hydrothermal activity occur at tectonic plate boundaries, along rifts, and at hot spots. These are the places where most mineral resources form. Tectonic effects are then overprinted by the influence of the hydrologic cycle.

Methods of Mineral Exploration

Copper Staining



Originally, exploration entailed looking for a "*show*" of minerals in a hillside, cliff-face, etc. This "show" would include quartz veins or oxidative staining that included the element(s) of interest.

The rock was then *assayed* (analyzed) to see what the concentration of the element of interest was.

A claim could be "*staked*" by simply staking off the area.

These days, laws are more stringent and exploration techniques are more sophisticated.

Methods of Mineral Exploration

Geophysics: gravity, magnetism. Also radioactivity. **GRAVITY**

Force of gravity increases between 2 objects with an increase in mass of either one.

Use a gravity meter to explore local variations in rock density: (mass = density x volume).







Neaker

Methods of Mineral Exploration



Methods of Mineral Exploration





Positive Magnetic Anomaly: magnetic field strength above the ₂₅ average – vice versa for a negative anomaly.

Methods of Mineral Exploration

Reddish areas are positive anomalies; Bluish areas are negative anomalies.



Magnetic Anomaly Map

Methods of Mineral Exploration

Geochemical prospecting – looking at concentrations of metals in the weathered rock (soil) over a wide area.

Plants can also be analyzed.

Even soil gases and groundwater can be used.



Groundwater Sampling for Mineral Exploration

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Ore Exploration and Production



Core drilling helps determine the extent of an ore deposit.

Mining

The type of mining depends on the proximity of the ore body to the surface.

Surface Mining

Most suitable for large ore bodies and for low grade, disseminated ore bodies.

Usually ore bodies are mined by a combination of surface and underground methods.

Initial mining is by surface methods, but as mining deepens, the amount of waste to be removed increases - soon economical to go underground.

Surface mining requires huge amounts of material to be removed – highly mechanized.



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Mining

Surface Mining (cont.)

Two types: *open pit* (massive ore bodies) & *strip mining* (bedded ore bodies).

Open Pit – waste continues to be stripped off during operation. Widen the area to ensure slope stability as pit deepens.



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Mining

Surface Mining (cont.)

Strip Mining –

overburden is stripped off (dumped as **spoil banks** or **spoil piles**), the ore removed, and waste replaces it in a cyclic manner.

The waste fraction is almost constant throughout mine operation.



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Bingham Canyon, Utah Open-Pit Mine

Surface Mining





Adit = horizontal entry; Shaft = vertical entry.

Deepest mine = 3.5 km; Temp. ~55°C! Columns of material left for roof support. Rock bursts = rock falls due to pressure.

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Nonmetallic Mineral Resources

Architects and Geologists have a similar, but different language: Architects "Stone" = Geologists "Rock"; Architects "Marble": Any polished carbonate rock; Architects "Granite": Any rock containing quartz and feldspar regardless of whether it is igneous or metamorphic.

Dimension Stone: intact

slabs/blocks of rock. A "Quarry" provides "stone", whereas a "Mine" provides "ore". Split vertically with wedges, split horizontally with wireline saw, thermal lance (blow torch), or high pressure water jet.





Nonmetallic Rock and Mineral Resources



TABLE 15.2 Common Nonmetallic Resources

Limestone	Sedimentary rock made of calcite; used for gravel or cement
Crushed stone	Any variety of coherent rock (limestone, quartzite, granite, gneiss)
Siltstone	Beds of sedimentary rock; used to make flagstone
Granite	Coarse igneous rock; used for dimension stone
Marble	Metamorphosed limestone; used for dimension stone
Slate	Metamorphosed shale; used for roofing shingles
Gypsum	A sulfate salt precipitated from salt water: used for wallboard
Phosphate	From the mineral apatite; used for fertilizer
Pumice	Frothy volcanic rock; used to decorate gardens and paths
Clay	Very fine mica-like mineral in sediment; used to make bricks or pottery
Sand	From sandstone, beaches, or riverbeds; quartz sand is used for construction and for making glass
Salt	From the mineral halite, formed by evaporating saltwater; used for food seasoning, and for melting ice on roads
Sulfur	Occurs either as native sulfur, typically above salt domes, or in sulfide minerals; used for fertilizer and chemicals

Nonmetallic Mineral Resources

Activities in the U.S.



Nonmetallic Mineral Resources

Crushed Stone & Concrete

Crushed Stone: substrate for roads and railways, raw material for manufacturing cement, asphalt, etc.

Quarried by blasting and crushing.

Mortar: substance that holds bricks together.

Concrete: substance used for roads, walls, etc. Both contain **Cement**: ~66% lime (CaO); ~25% silica (SiO₂), ~9% Al₂O₃

and Fe_2O_3 (from clay or shale).

The ideal limestone contains these elements in those proportions. This is from the Jurassic Portland Limestone from England. Cement is now predominantly made by mixing limestone, clay, and quartz (sand) in the right proportions.

These are then heated in a kiln to ~1,450°C, which releases CO_2 and produces "clinker". This is crushed to give cement.

Nonmetallic Mineral Resources





In the Home						
Bricks (clay);	Per Cap	oita Annual Usage				
Cement; Porcelain; Crockery; Glass (silica); Drywall (gypsum); Salt; Cutlery (steel); Cookware; Countertops; Electrical Wire; Appliances	4,100 kg 3,860 kg 3,050 kg 2,650 kg 1,900 kg 550 kg 360 kg 220 kg 200 kg 140 kg 25 kg 10 kg	Stone Sand & Gravel Petroleum Coal Natural Gas Iron & Steel Cement Clay Salt Phosphate Aluminium Copper				
What Else???	5 kg	Zinc 39				

Mineral/Rock Resources are Finite

Lifetime (years) of Known Ore Resources

Metal	World Resource	U.S. Resources	
Iron	120	40	
Aluminium	330	2	
Copper	65	40	Strategic Metals : Mn (100%);
Lead	20	40	Pt (92%); Cr (73%); Co (95%).
Zinc	30	25	Percentages = amount needed
Gold	30	20	to be imported annually.
Platinum	45	1	
Nickel	75	<1	Needed for national security
Cobalt	50	<1	but not produced by host
Manganese	70	0	country.
Chromium	75	0	О Т

Environmental Effects

- 1) Waste dumps leaching and landslides
- 2) Subsidence/Earthquakes
- 3) Acid mine drainage
- **4) Pollution**



Mining & The Environment



Environmental Impacts of Mining



Environmental Impacts of Mining



Sulfides react with water and O_2 to form sulfuric acid (H₂SO₄).



Environmental Impacts of Mining



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Cutaway showing artist's concept of the complex of underground tunnels into which waste would be emplaced. A repository at Yucca Mountain would rely on the semiarid climate, natural barriers, and engineered barriers to contain and isolate waste for thousands of years.

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Summary



Summary

Resources & Reserves.

Mineral Deposits: Metallic, Nonmetallic.

Metallic Deposits: Native Metals; Smelting; Slag.

Ore Deposits: Ore; Grade; Gangue; Magmatic Deposits; Hydrothermal Deposits (Massive Sulfide, Vein, Porphyry Copper, Disseminated, Black Smoker); Secondary (Supergene) Enrichment Deposits; Mississippi-Valley-Type Deposits; Sedimentary Deposits (BIFs, Evaporite, Mn-Nodules); Residual Mineral Deposits; Placer Deposits.

Distribution of Ore Deposits.

- **Mineral Exploration**: Show, Staining, Veins; Gravity; Magnetism; Groundwater.
- Mining: Open Pit, Strip, Underground (adit, shaft).
- **NonMetallic Resources**: Architects vs. Geologists; Dimension Stone; Crushed Stone; Concrete (Mortar, Cement).
- Mining and The Environment: Subsidence; Waste Dumps; Acid Mine Drainage; Nuclear Waste.