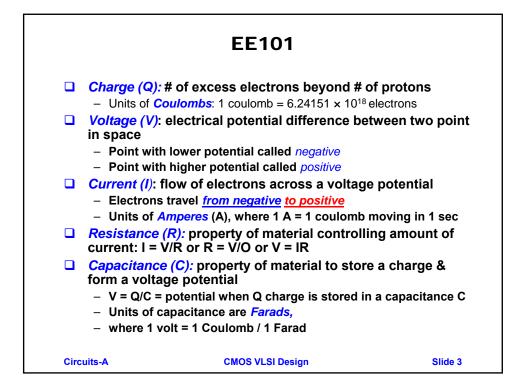
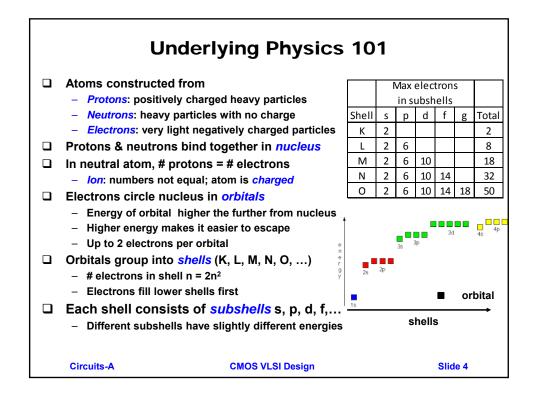
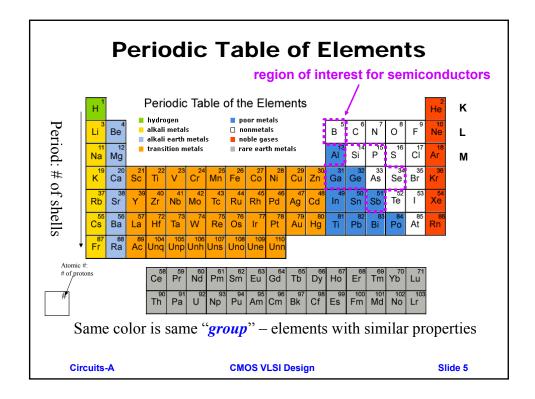
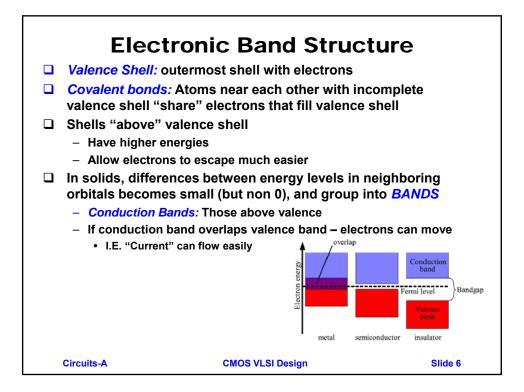
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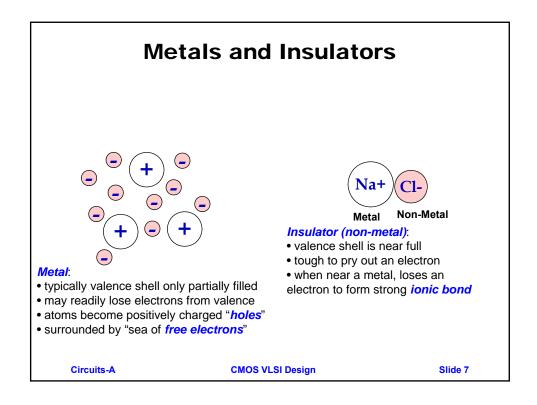
	Οι	Itline: Circuits	
	Lecture A		
	– Physics 101		
	- Semiconductors	for Dummies	
	– CMOS Transisto	rs for logic designers	
	Lecture B		
	 NMOS Logic 		
	 CMOS Inverter ar 	nd NAND Gate Operation	
	 CMOS Gate Desi 	gn	
	 Adders 		
	 Multipliers 		
	Lecture C		
	 Pass Transistors 		
	 Tri-states 		
	 Multiplexors 		
	 Latches 		
	 FlipFlops 		
	 Barrel Shifters 		
Circ	cuits-A	CMOS VLSI Design	Slide 2



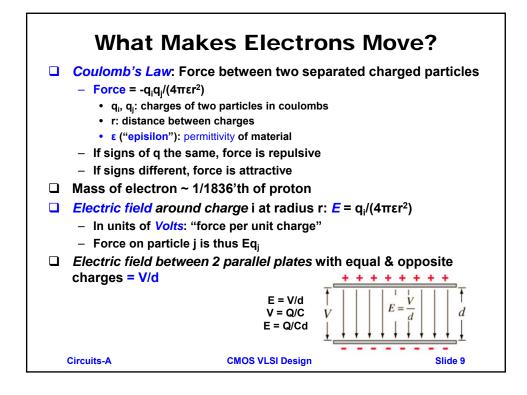


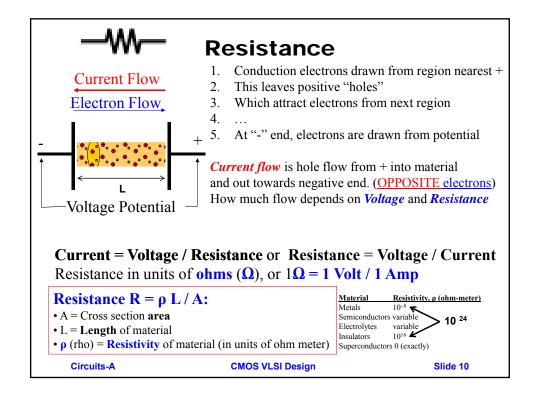


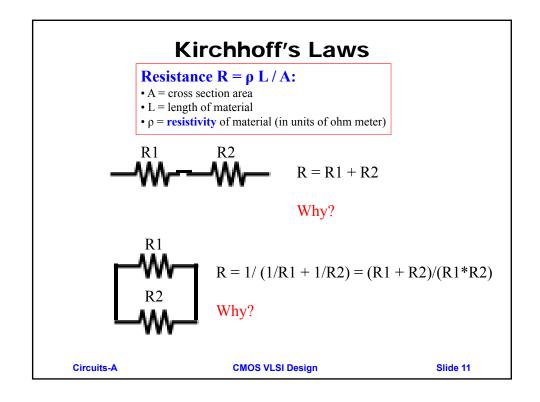


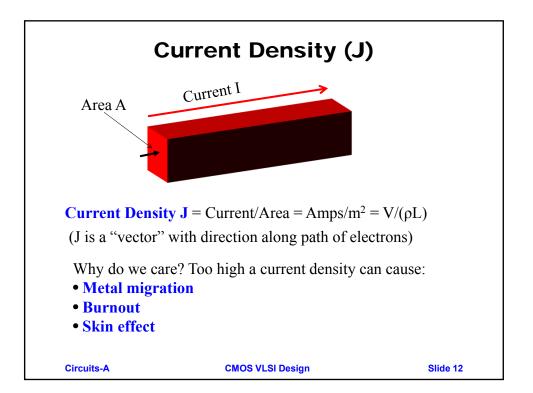


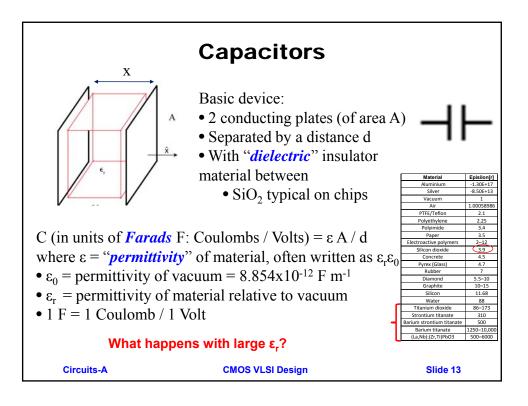
Element	Symbol	Atomic #	Valence Shell	Electrons in Valence	Electrons to fill valence	
Aluminum	AI	13	М	3	15	
Copper	Cu	29	Ν	1	31	
Gold	Au	79	0	19	31	
Boron	В	5	L	3	5	
Phosphorus	Ρ	15	М	5	13	
Silicon	Si	14	М	4	14	
Germanium	Ge	32	Ν	4	28	

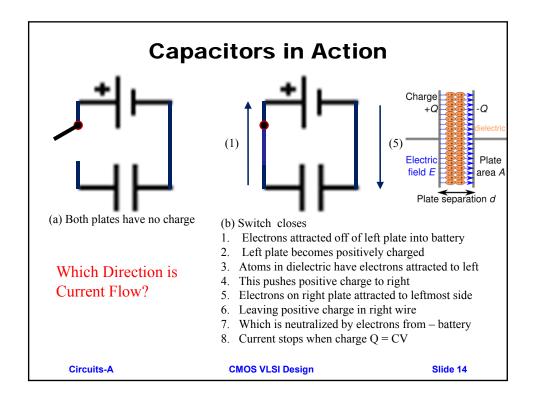


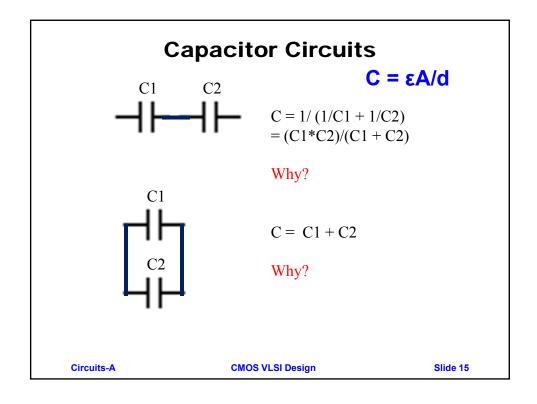


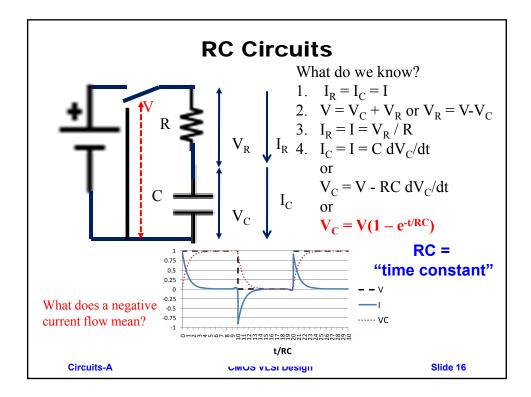


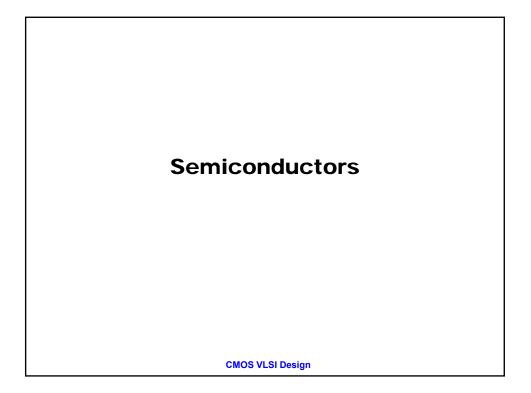












			K	(e	y	Μ	at	te	ria	als	S:	S	İ,	B	, F)		-
H		Periodic Table of the Elements												He				
3 Li	Be							onmet	als			B	C	N ⁷	08	F ⁹	¹⁰ Ne	
11 Na	12 Mg							-	gases I. arth metals				14 Si	15 P	5 S	CI CI	18 Ar	
¹⁹ К	Ca ²⁰	SC ²¹	Ti Ti	V ²³	Cr ²⁴	25 Mn	Fe ²⁶	C0	28 Ni	Cu Cu	Zn ³⁰	31 Ga	Ge ³²	As	34 Se	Br Br	Kr ³⁶	
37 Rb	³⁸ Sr	39 Y	Zr Zr	41 Nb	42 Mo	43 TC	44 Ru	45 Rh	46 Pd	47 Ag	48 Cd	49 In	50 Sn	51 Sb	Te ⁵²	1 ⁵³	Xe ⁵⁴	
Cs	56 Ba	57 La	72 Hf	73 Ta	W ⁷⁴	75 Re	76 Os	77 Ir	Pt	79 Au	80 Hg	81 Ti	82 Pb	83 Bi	84 Po	At 85	86 Rn	
87 Fr	⁸⁸ Ra	89 Ac	¹⁰⁴ Unq	Unp	106 Unh	¹⁰⁷ Uns	¹⁰⁸ Uno	Une	Unn									
	Material					Atomic # Electrons p						ber	She	II				
	S	Silicon Si			1	4			2, 8, 4									
	В	Boron B Phosphorus				5 15			2, 3									
	Ρ								2, 8, <mark>5</mark>									
Circu	its-A							смо	OS VL	SI D	esigr	1						Slide 18

