AME 538 Homework 23 Due: Friday, 3 November 2000, in class

1. Given that a shock wave is propagating at $D = 500 \ m/s$ into N_2 at rest with $T_1 = 130 \ K$ and $P_1 = 2 \ MPa$, and that the diatomic nitrogen is well modeled by a van der Waals thermal and caloric state equation

$$P = \frac{\rho RT}{1 - b\rho} - a\rho^2,$$
$$e = e_o + c_v (T - T_o) + a (\rho_o - \rho),$$

find appropriate values for the constants in a thermodyanics text, and solve the Rankine-Hugoniot shock jump equations to describe the shocked state, P_2, T_2, ρ_2, u_2 .