

**Resolved Viscous Detonation in $H_2/O_2/Ar$ Using
Intrinsic Low Dimensional Manifolds and Wavelet
Adaptive Multilevel Representation**

by

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presented to the

Second International Workshop on Combustion Modeling

Instituto Tecnológico de Veracruz

22 February 2001

Support: NSF, AFOSR, LANL

Acknowledgments

Prof. Samuel Paolucci, Faculty, ND-AME,

Mr. Sandeep Singh, Ph.D. Candidate, ND-AME,

Mr. Yevgenii Rastigejev, Ph.D. Candidate, ND-AME.

Outline

- Motivation for Intrinsic Low Dimensional Manifold (ILDM) technique (Maas & Pope, 1992) for spatially homogeneous systems (ODEs, no convection-diffusion)
- Compressible Reactive Navier-Stokes equations
- Details of ILDM implementation in operator splitting method
- Wavelet Adaptive Multilevel Representation (WAMR) technique (Paolucci & Vasilyev) for spatial discretization
- Results for one-dimensional viscous $H_2/O_2/Ar$ detonation with detailed kinetics (Singh, Rastigejev, Paolucci, and Powers, *Combustion Theory and Modeling*, submitted 2001) using operator splitting for convection-diffusion
- Work in progress: systematic, center manifold-motivated correction for convection-diffusion using ILDMs
- Conclusions

Some Important Questions

- Do we have resolved, accurate solutions for systems with detailed kinetics?
- How can ILDM improve the calculation of systems with detailed kinetics?
- How can ILDM, derived for spatially homogeneous systems, be used rationally in systems in which convection and diffusion are important?

Motivation for ILDM

- Detailed finite rate kinetics critical in reactive fluid mechanics
- Common detailed kinetic models are computationally expensive.
- Expense increases with
 - number of species and reactions modeled (linear effect),
 - *stiffness*–ratio of slow to fast time scales, (geometric effect).
- chemical time scales typically more demanding than convection-diffusion
- Reduced kinetics necessary given current computational resources.

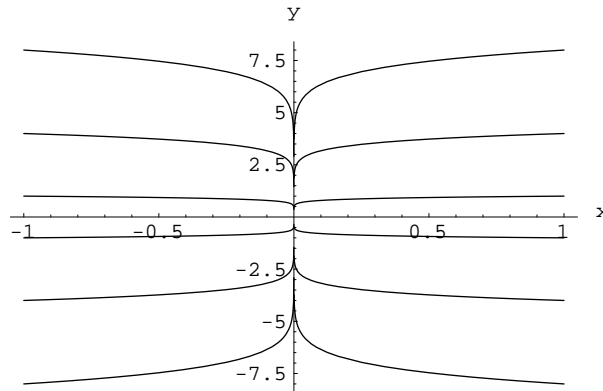
Intrinsic Low-Dimensional Manifold Method (ILDMM)

- Uses a dynamical systems approach,
- Most appropriate for spatially homogeneous systems (ODEs)
- Does not require imposition of *ad hoc* partial equilibrium or steady state assumptions,
- Fast time scale phenomena are systematically equilibrated,
- Slow time scale phenomena are resolved in time,
- Computation time reduced by factor of ~ 3 for non-trivial combustion problem considered here; manifold gives much better roadmap to find solution relative to general implicit solution techniques (Singh, *et al.* 2001)
- Speed up factor depends on
 - initial conditions,
 - stiffness ratio
 - dimension of ILDM

Simplest Example

$$\begin{aligned}\frac{dx}{dt} &= -10x, & x(0) &= x_o, \\ \frac{dy}{dt} &= -y, & y(0) &= y_o.\end{aligned}$$

- Stable equilibrium at $(x, y) = (0, 0)$; stiffness ratio = 10.
- ILDM is $x = 0$



- Parameterization of manifold: $x(s) = 0; y(s) = s$.

$$\frac{dy}{dt} = \frac{dy}{ds} \frac{ds}{dt}, \quad \text{chain rule}$$

$$-y(s) = \frac{dy}{ds} \frac{ds}{dt}, \quad \text{substitute from ODE and manifold}$$

$$-s = (1) \frac{ds}{dt}, \quad \text{no longer stiff!}$$

$$s = s_o e^{-t},$$

$$x(t) = 0; \quad y(t) = s_o e^{-t}.$$

- Projection onto manifold for s_o , induces small phase error.

Compressible Reactive Navier-Stokes Equations

$$\begin{aligned} \frac{\partial \rho}{\partial t} + \frac{\partial}{\partial x}(\rho u) &= 0, & \text{mass} \\ \frac{\partial}{\partial t}(\rho u) + \frac{\partial}{\partial x}(\rho u^2 + p - \tau) &= 0, & \text{momentum} \\ \frac{\partial}{\partial t}\left(\rho\left(e + \frac{u^2}{2}\right)\right) + \frac{\partial}{\partial x}\left(\rho u\left(e + \frac{u^2}{2}\right) + u(p - \tau) + q\right) &= 0, & \text{energy} \\ \frac{\partial}{\partial t}(\rho y_l) + \frac{\partial}{\partial x}(\rho u y_l + j_l) &= 0, & (l = 1, \dots, L-1), \quad \text{elements} \\ \frac{\partial}{\partial t}(\rho Y_i) + \frac{\partial}{\partial x}(\rho u Y_i + J_i^m) &= \dot{\omega}_i M_i, & (i = 1, \dots, N-L), \quad \text{species} \\ \tau &= \frac{4}{3}\mu \frac{\partial u}{\partial x}, & \text{Newtonian gas with Stokes' assumption} \\ J^q &= -k \frac{\partial T}{\partial x} + \sum_{i=1}^N J_i^m \left(h_i^o + \int_{T_o}^T c_{pi}(\hat{T}) d\hat{T} \right) - \Re T \sum_{i=1}^N \frac{\mathcal{D}_i^T}{M_i} \left(\frac{1}{\chi_i} \frac{\partial \chi_i}{\partial x} + \left(1 - \frac{M_i}{M}\right) \frac{1}{p} \frac{\partial p}{\partial x} \right), & \text{Fourier's law} \\ J_i^m &= \rho \sum_{j=1, j \neq i}^N \frac{M_i}{M} Y_j \mathcal{D}_{ij} \left(\frac{1}{\chi_j} \frac{\partial \chi_j}{\partial x} + \left(1 - \frac{M_j}{M}\right) \frac{1}{p} \frac{\partial p}{\partial x} \right) - \mathcal{D}_i^T \frac{1}{T} \frac{\partial T}{\partial x}, & (i = 1, \dots, N), \quad \text{Fick's law} \\ y_l &= m_l \sum_{i=1}^N \frac{\phi_{il}}{M_i} Y_i, & (l = 1, \dots, L-1), \quad \text{element mass fraction} \\ M &= \sum_{i=1}^N M_i \chi_i, & \text{mean molecular mass} \\ \chi_i &= \frac{M}{M_i} Y_i, & (i = 1, \dots, N), \quad \text{mole fraction} \\ j_l &= m_l \sum_{i=1}^N \frac{\phi_{il}}{M_i} J_i, & (l = 1, \dots, L-1), \quad \text{element mass flux} \\ \sum_{i=1}^N Y_i &= 1, & \text{mass fraction constraint} \\ \sum_{l=1}^L y_l &= 1, & \text{element mass fraction constraint} \\ \dot{\omega}_i &= \sum_{j=1}^J a_j T^{\beta_j} \exp\left(\frac{-E_j}{\Re T}\right) (\nu''_{ij} - \nu'_{ij}) \prod_{k=1}^N \left(\frac{\rho Y_k}{M_k}\right)^{\nu'_{kj}}, & (i = 1, \dots, N-L) \quad \text{law of mass action} \\ p &= \rho \Re T \sum_{i=1}^N \frac{Y_i}{M_i}, & \text{thermal equation of state} \\ e &= \sum_{i=1}^N Y_i \left(h_i^o + \int_{T_o}^T c_{pi}(\hat{T}) d\hat{T} - \frac{\Re T}{M_i} \right). & \text{caloric equation of state} \end{aligned}$$

N species, L elements, J reactions

$4N + L + 7$ equations in $4N + L + 7$ unknowns

Focus on element conservation

- $L - 1$ element equations formed; $N - L$ species equations, instead of the typical $N - 1$ species equations,
- facilitates a proper use of ILDM in upcoming operator splitting,
- In general element mass fractions change due to mass diffusion

$$\rho \frac{dy_l}{dt} = -\frac{\partial j_l}{\partial x}.$$

- Assuming
 - Soret and DuFour effects are negligible, $D_i^T = 0$,
 - Deviations of molecular mass not large $M_i \sim M$,
 - Diffusion coefficients are equal, $\mathcal{D}_{ij} \sim \mathcal{D}$,
- Molecular mass diffusion reduces to $J_i^m = -\rho \mathcal{D} \frac{\partial Y_i}{\partial x}$.
- Elemental mass diffusion reduces to $j_l = -\rho \mathcal{D} \frac{\partial y_l}{\partial x}$.
- Evolution equation for element mass fraction becomes

$$\rho \frac{dy_l}{dt} = \mathcal{D} \frac{\partial}{\partial x} \left(\rho \frac{\partial y_l}{\partial x} \right).$$

- In uniformly premixed problem with no boundary influences then, all element concentrations are constant for all time:

$$\frac{dy_l}{dt} = 0.$$

Operator Splitting Technique for Convection-Diffusion

- Equations are of form

$$\frac{\partial}{\partial t} \mathbf{q}(x, t) + \frac{\partial}{\partial x} \mathbf{f}(\mathbf{q}(x, t)) = \mathbf{g}(\mathbf{q}(x, t)), \quad \mathbf{q}, \mathbf{f}, \mathbf{g} \in \mathfrak{R}^{N+2}.$$

where

$$\mathbf{q} = \left(\rho, \rho u, \rho \left(e + \frac{u^2}{2} \right), \rho y_l, \rho Y_i \right)^T.$$

- \mathbf{f} models convection-diffusion
- \mathbf{g} models reaction source terms
- Splitting

1. Inert convection-diffusion step:

$$\begin{aligned} \frac{\partial}{\partial t} \mathbf{q}(x, t) + \frac{\partial}{\partial x} \mathbf{f}(\mathbf{q}(x, t)) &= 0, \\ \frac{d}{dt} \mathbf{q}_i(t) &= -\Delta_x \mathbf{f}(\mathbf{q}_i(t)). \end{aligned}$$

Δ_x is any spatial discretization operator, here a wavelet operator.

2. Reaction source term step:

$$\begin{aligned} \frac{\partial}{\partial t} \mathbf{q}(x, t) &= \mathbf{g}(\mathbf{q}(x, t)), \\ \frac{d}{dt} \mathbf{q}_i(t) &= \mathbf{g}(\mathbf{q}_i(t)). \end{aligned}$$

- Operator splitting with implicit stiff source solution can induce non-physical wave speeds! (LeVeque and Yee, *JCP* 1990)

ILDm Implementation in Operator Splitting

- Form of equations in source term step:

$$\frac{d}{dt} \begin{pmatrix} \rho \\ \rho u \\ \rho \left(e + \frac{u^2}{2} \right) \\ \rho y_l \\ \rho Y_i \end{pmatrix} = \begin{pmatrix} 0 \\ 0 \\ 0 \\ 0 \\ \dot{\omega}_i M_i \end{pmatrix}.$$

$$l = 1, \dots, L - 1, \quad i = 1, \dots, N - L.$$

- Equations reduce to

$$\rho = \rho_o, \quad u = u_o, \quad e = e_o, \quad y_l = y_{lo},$$

$$\frac{dY_i}{dt} = \frac{\dot{\omega}_i M_i}{\rho_o}, \quad i = 1, \dots, N - L$$

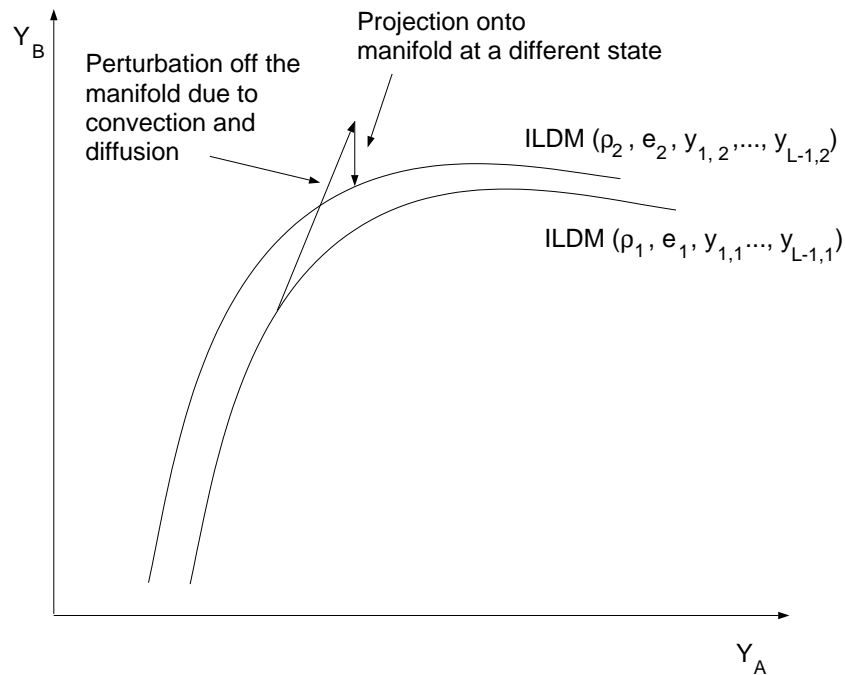
- $\dot{\omega}_i$ has dependency on ρ , e , y_l , and Y_i
- ODEs for Y_i are stiff, usually solved with implicit methods.
- ODEs for Y_i can be attacked with manifold methods to remove stiffness with ILDM with ρ , e , y_l, \dots, y_{L-1} parameterization.

Necessary Dimension of ILDM

- Spatial discretization of PDEs results in a set of adiabatic, isochoric spatially homogeneous reactors,
- N species with L elements at constant e and ρ gives rise to a $(N - L)$ -dimensional phase space (same as composition space),
- To resolve M slow time scales, we identify M -dimensional subspaces (manifolds), $M < (N - L)$, embedded within the $(N - L)$ -dimensional phase space on which the M slow time scale events evolve,
 - Fast time scale events rapidly move to the manifold,
 - Slow time scale events move on the manifold,
 - Because of convection-diffusion, e , ρ , y_l vary, requiring a $K = M + L + 1$ -dimensional manifold.
 - If y_l conserved (premixed with no preferential diffusion), dimension of manifold is reduced by $L - 1$.
 - e.g., for $M = 1$ in premixed $H_2/O_2/Ar$ with no preferential diffusion, small molecular mass deviation, and negligible Soret/DuFour diffusion, we need $K = 3$.

Implementation of ILDMs with convection-diffusion and operator splitting

- To minimize phase error, must integrate full equations until sufficiently close to ILDM
- When near ILDM, M slow equations are integrated, other variables found by table lookup
- Convection-diffusion step *applied to all variables* perturbs system from ILDM
- In next reaction step, project to ILDM at *different* value of ρ , e , y_1, \dots, y_{N-1} .



Formulation of General ILDMs

- A spatially homogeneous adiabatic, isochoric chemically reactive system of N species in L elements is modeled by a set of non-linear ordinary differential equations:

$$\frac{d\mathbf{x}}{dt} = \mathbf{F}(\mathbf{x}), \quad \mathbf{x}(0) = \mathbf{x}_o,$$

\mathbf{x} : species concentration; $\mathbf{x} \in \mathfrak{R}^{N-L}$

- Equilibrium points defined by

$$\mathbf{x} = \mathbf{x}_{eq} \text{ such that } \mathbf{F}(\mathbf{x}_{eq}) = 0.$$

- Consider a system near equilibrium (the argument can and must be extended for systems away from equilibrium) with $\tilde{\mathbf{x}} = \mathbf{x} - \mathbf{x}_{eq}$.

- Linearization gives

$$\frac{d\tilde{\mathbf{x}}}{dt} = \mathbf{F}_{\mathbf{x}} \cdot \tilde{\mathbf{x}},$$

where $\mathbf{F}_{\mathbf{x}}$ is a *constant* Jacobian matrix.

- Schur decompose the Jacobian matrix:

$$\mathbf{F}_{\mathbf{x}} = \mathbf{Q} \cdot \mathbf{U} \cdot \mathbf{Q}^T$$

$$\mathbf{Q} = \begin{pmatrix} \vdots & \vdots & & \vdots \\ q_1 & q_2 & \cdots & q_{N-L} \\ \vdots & \vdots & & \vdots \end{pmatrix}, \quad \mathbf{U} = \begin{pmatrix} \lambda_1 & u_{1,2} & \cdots & u_{1,N-L} \\ 0 & \lambda_2 & \cdots & u_{2,N-L} \\ 0 & \cdots & \ddots & \vdots \\ 0 & \cdots & 0 & \lambda_{N-L} \end{pmatrix}, \quad \mathbf{Q}^T = \begin{pmatrix} \cdots & q_1^T & \cdots \\ \cdots & q_2^T & \cdots \\ \vdots & \vdots & \vdots \\ \cdots & q_{N-L}^T & \cdots \end{pmatrix}$$

Formulation of General Manifolds (cont.)

- \mathbf{Q} is an orthogonal matrix with real Schur vectors q_i in its columns.
- \mathbf{U} is an upper triangular matrix with eigenvalues of $\mathbf{F}_{\mathbf{x}}$ on its diagonal, sometimes placed in order of decreasing magnitude.
- The Schur vectors q_i form an orthonormal basis which spans the phase space, \mathfrak{R}^{N-L} .
- We then define M slow time scales, $M < N - L$.
- Next define a non-square matrix \mathbf{W} which has in its rows the Schur vectors associated with the fast time scales:

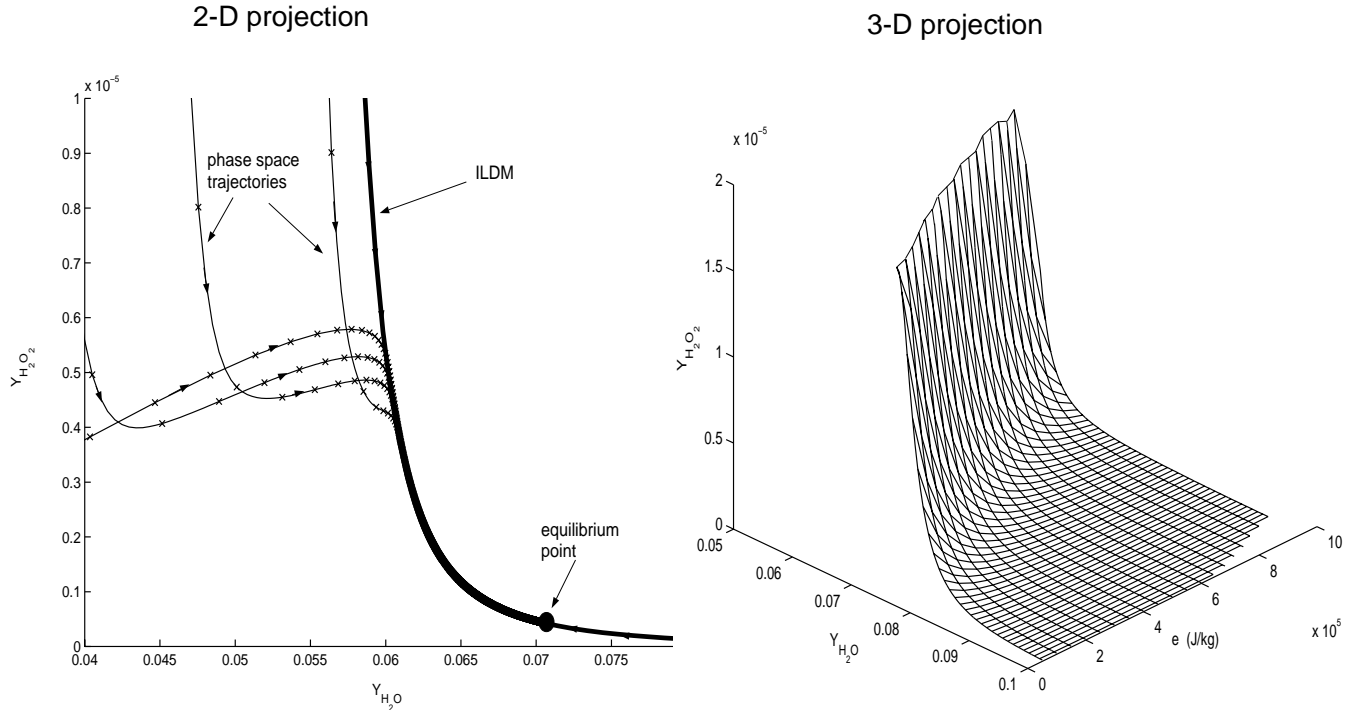
$$\mathbf{W} = \begin{pmatrix} \cdots & \cdots & q_{M+1}^T & \cdots & \cdots \\ \cdots & \cdots & q_{M+2}^T & \cdots & \cdots \\ & & \vdots & & \\ \cdots & \cdots & q_{N-L}^T & \cdots & \cdots \end{pmatrix}.$$

- Letting the fast time scale events equilibrate defines the manifold:

$$\mathbf{W} \cdot \mathbf{F}(\mathbf{x}) = 0.$$

Sample ILDM for $H_2/O_2/Ar$

- Based on $N = 9$, $J = 37$ mechanism of Maas and Warnatz,
- Projection in Y_{H_2O} , $Y_{H_2O_2}$ plane and Y_{H_2O} , $Y_{H_2O_2}$, e space
- Adiabatic ($e = 8 \times 10^5 \text{ J/kg}$), isochoric ($\rho = 5.0 \times 10^{-4} \text{ kg/m}^3$),
 $y_H = 0.01277$, $y_O = 0.10137$, $y_{Ar} = 0.88586$,
- We can get e.g. $p(\rho, e, Y_{H_2O})$, $T(\rho, e, Y_{H_2O})$, $Y_H(\rho, e, Y_{H_2O})$, \dots
- Linear interpolation used for points not in table,
- Captures $\sim 0.1 \mu s$ reaction events.



Wavelet Adaptive Multilevel Representation (WAMR) Technique

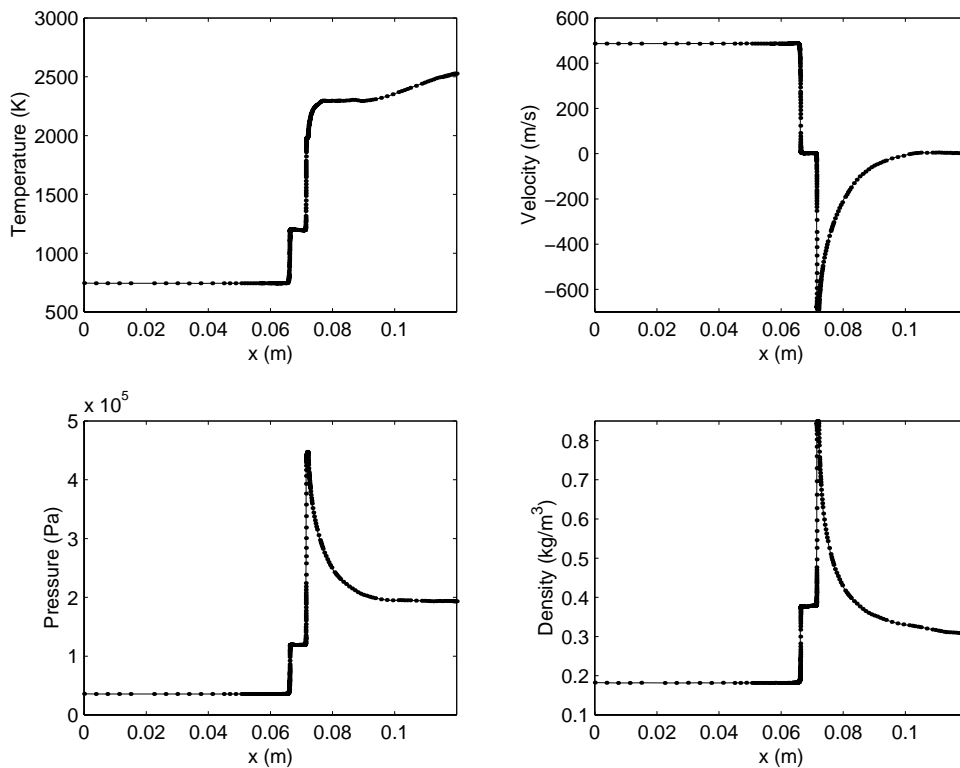
- Summary of standard spatial discretization techniques
 - Finite difference-good spatial localization, poor spectral localization, and slow convergence,
 - Finite element- good spatial localization, poor spectral localization, and slow convergence,
 - Spectral-good spectral localization, poor spatial localization, but fast convergence.
- Wavelet technique
 - See e.g. Vasilyev and Paolucci, “A Fast Adaptive Wavelet Collocation Algorithm for Multidimensional PDEs,” *J. Comp. Phys.*, 1997,
 - Basis functions have compact support,
 - Well-suited for problems with widely disparate spatial scales,
 - Good spatial and spectral localization, and fast (spectral) convergence,
 - Easy adaptable to steep gradients via adding collocation points,
 - Spatial adaptation is automatic and dynamic to achieve prescribed error tolerance.

Ignition Delay in Premixed $H_2/O_2/Ar$

- Consider standard problem of Fedkiw, Merriman, and Osher, *J. Comp. Phys.*, 1996,
- Shock tube with premixed H_2 , O_2 , and Ar in 2/1/7 molar ratio,
- Initial inert shock propagating in tube,
- Reaction commences shortly after reflection off end wall,
- Detonation soon develops,
- Model assumptions
 - One-dimensional,
 - Mass, momentum, and energy diffusion,
 - Nine species, thirty-seven reactions,
 - Ideal gases with variable specific heats.

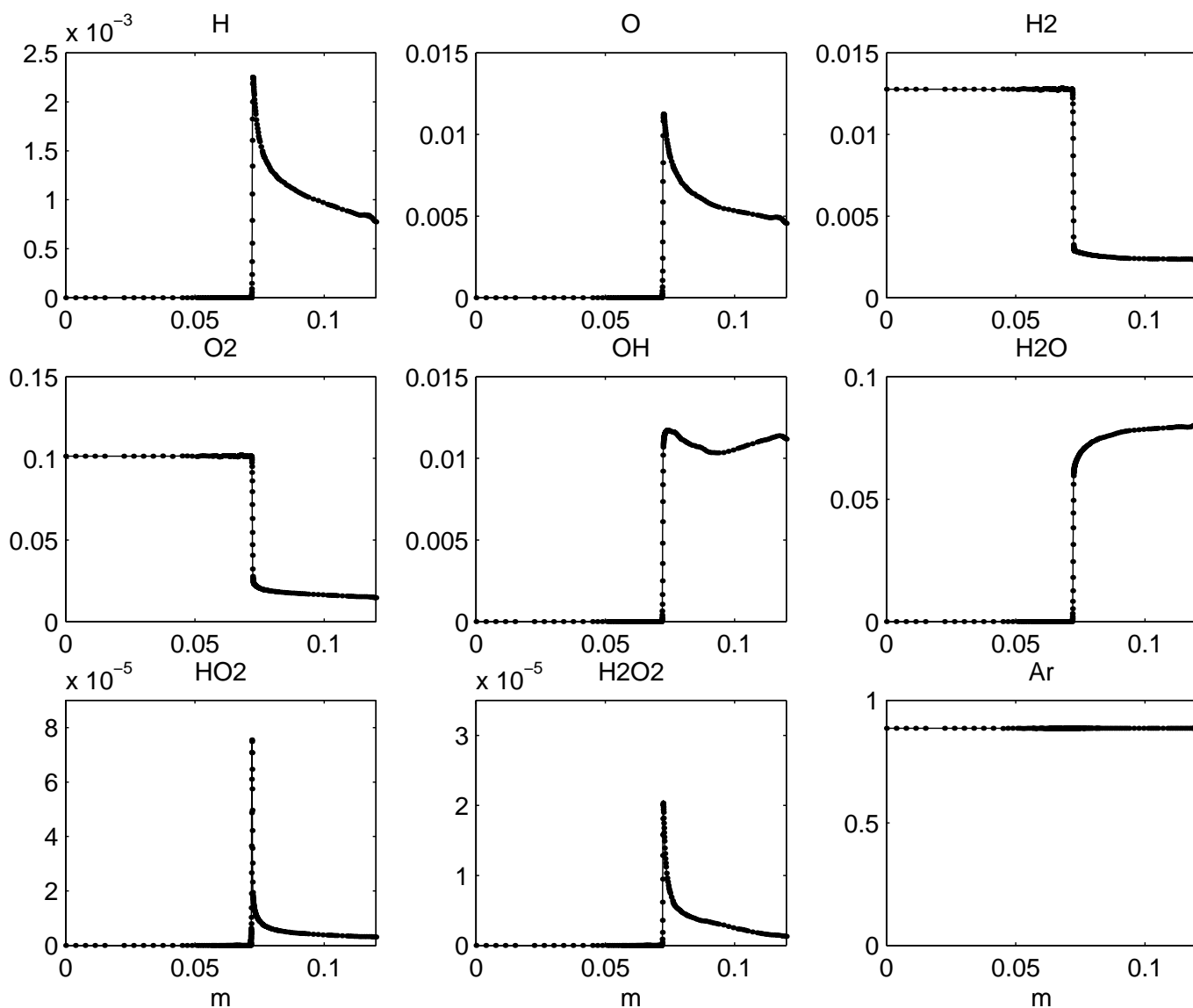
Viscous $H_2 - O_2$ Ignition Delay with Wavelets and ILDM

- $t = 195 \mu s$, 300 collocation points, 15 wavelet scale levels
- ILDM gives nearly identical results as full chemistry
- WAMR spatial discretization, implicit linear trapezoidal convection-diffusion time stepping, explicit (ILDM)/implicit (non-ILDM) reaction time stepping
- *Viscous shocks, inductions zones, and entropy layers spatially resolved!*



Viscous $H_2 - O_2$ Ignition Delay with Wavelets and ILDM

- $t = 195 \mu s$
- ILDM gives nearly identical results as full chemistry

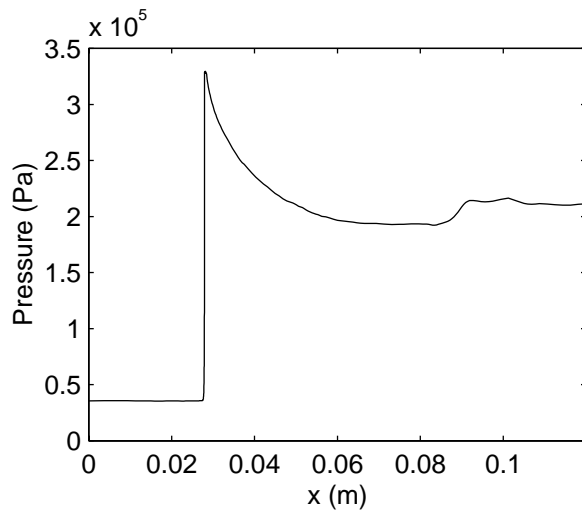


Viscous $H_2 - O_2$ Ignition Delay with Wavelets

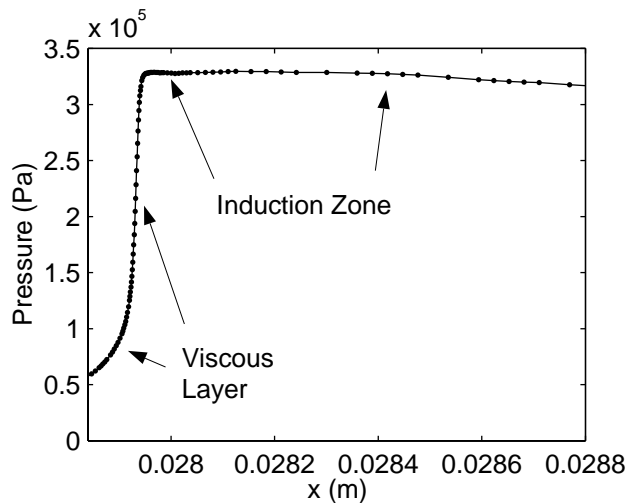
Global and Fine Scale Structures

- $t = 230 \mu s$, Induction zone length: $\sim 470 \mu m$, Viscous shock thickness: $\sim 50 \mu m$ (should use smaller μ),
- No significant reaction in viscous shock zone.

Global View

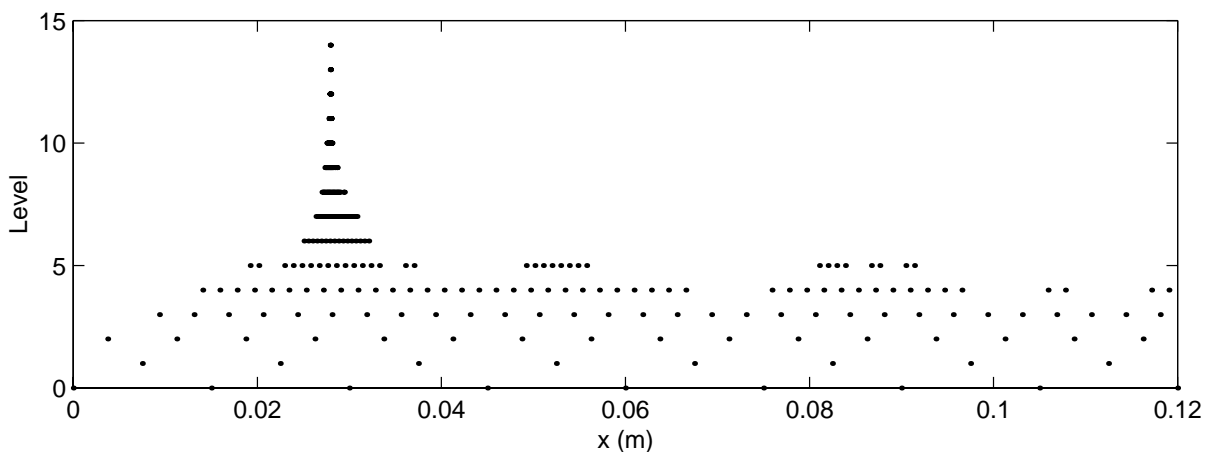
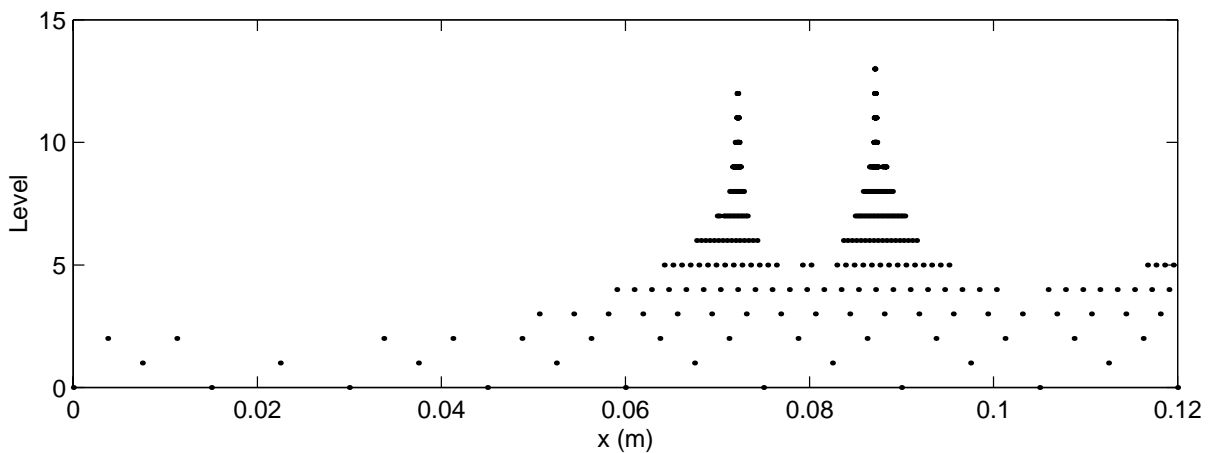


Fine Scale Structure



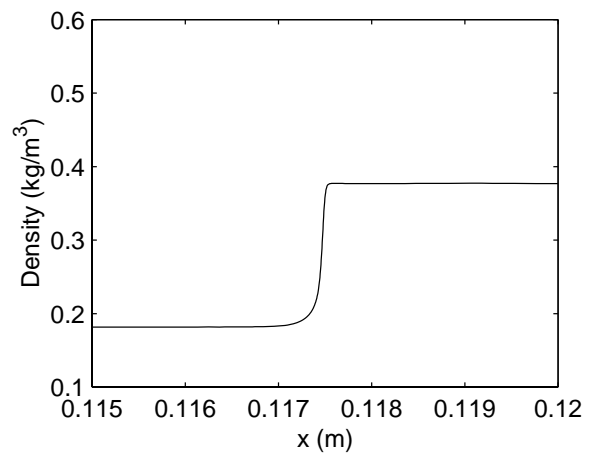
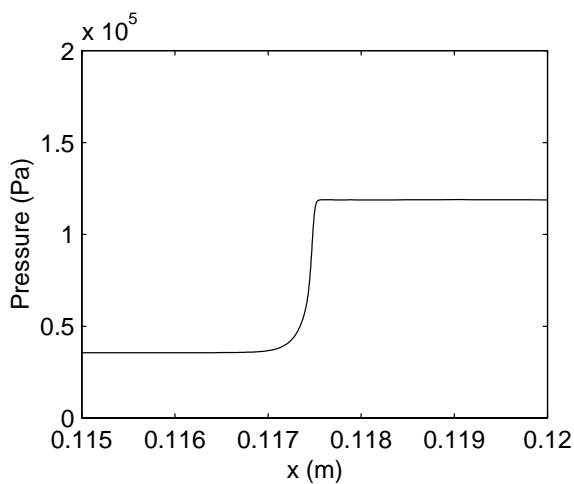
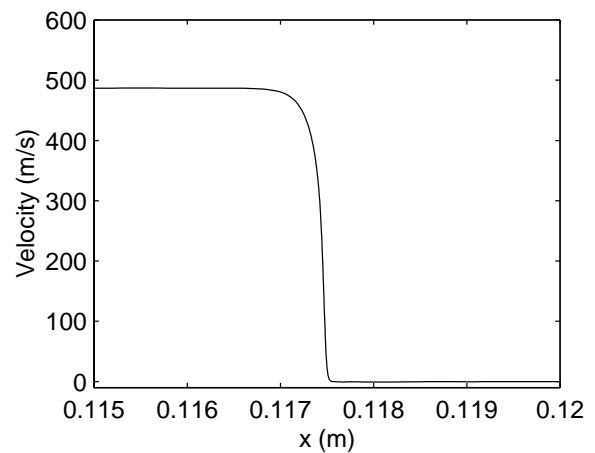
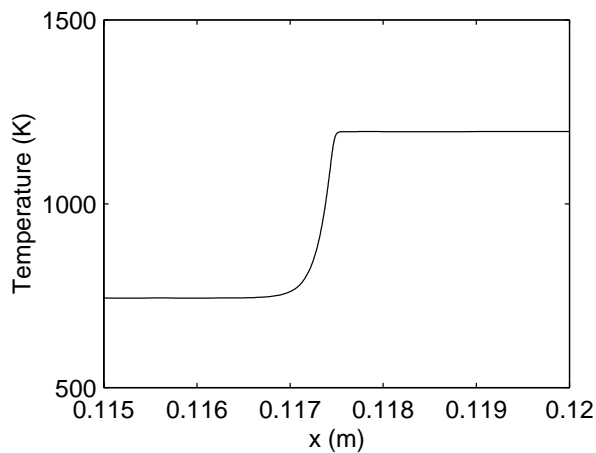
Viscous $H_2 - O_2$ Ignition Delay with Wavelets, Instantaneous Distributions of Collocation Points

- $t = 180 \mu s$, two-shock structure with consequent collocation point distribution,
- $t = 230 \mu s$, one-shock structure with evolved collocation point distribution.



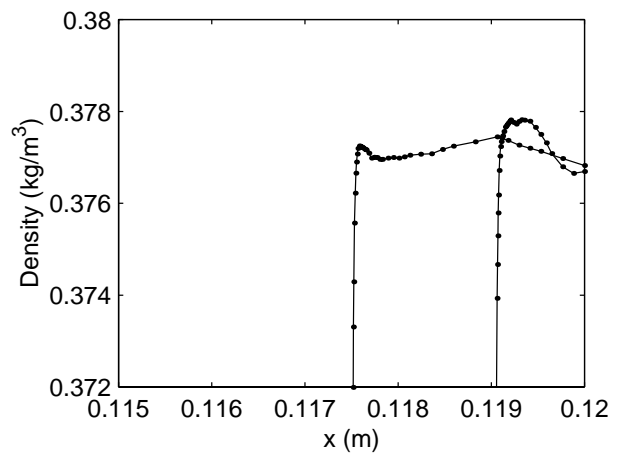
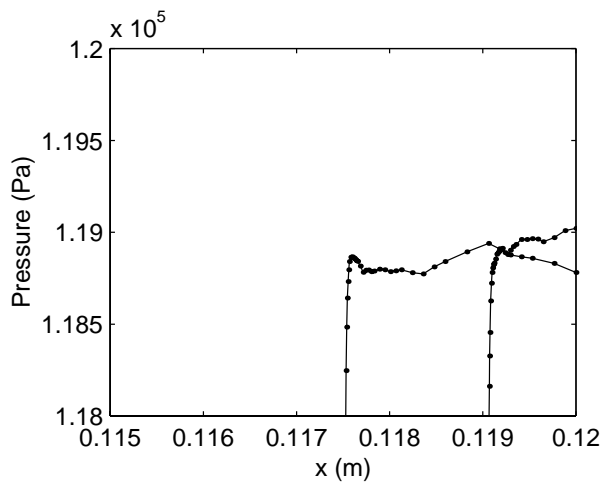
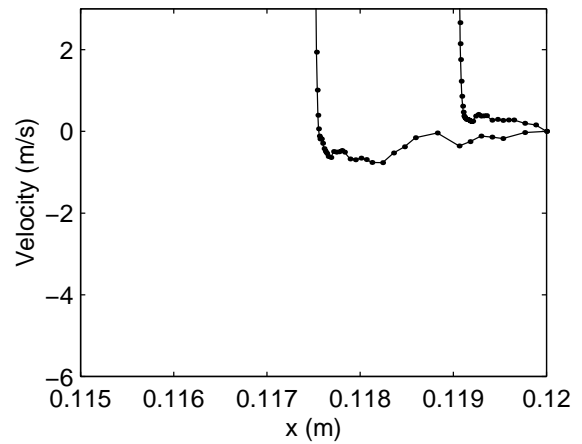
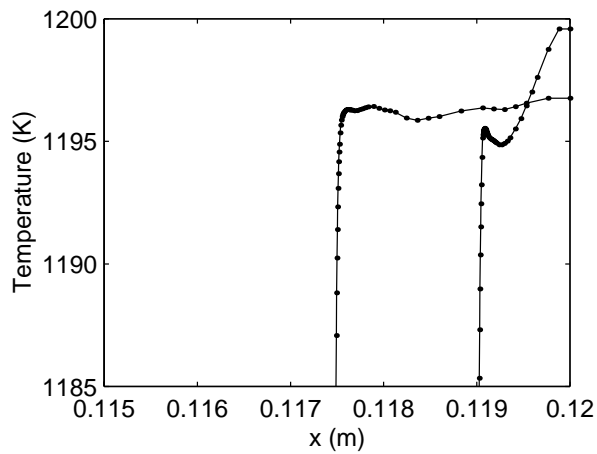
Post Reflection Entropy Layer?: Viscous Wavelet Results

- No significant entropy layer evident on macroscale after shock reflection when resolved viscous terms considered,
- Inviscid codes with coarse gridding introduce a larger entropy layer due to numerical diffusion,
- Unless suppressed, unphysically accelerates reaction rate.



Post Reflection Entropy Layer: Viscous Wavelet Results

- small entropy layer evident on finer scale,
- temperature rise $\sim 5\text{ K}$; dissipates quickly,
- inviscid calculations before adjustment give persistent temperature rise of $\sim 20\text{ K}$; reaction acceleration small.



Center manifold-motivated correction for small convection-diffusion

- Consider system of Davis and Skodje, 1999, extended for diffusion

$$\frac{\partial y_1}{\partial t} = \underbrace{-y_1}_{\text{reaction}} + \underbrace{\epsilon \frac{\partial^2 y_1}{\partial x^2}}_{\text{diffusion}},$$

$$\frac{\partial y_2}{\partial t} = \underbrace{-\gamma y_2 + \frac{(\gamma - 1)y_1 + \gamma y_1^2}{(1 + y_1)^2}}_{\text{reaction}} + \underbrace{\epsilon \frac{\partial^2 y_2}{\partial x^2}}_{\text{diffusion}},$$

$$y_1(x, 0) = x, \quad y_1(0, t) = 0, \quad y_1(1, t) = 1,$$

$$y_2(x, 0) = 0.55x \quad y_2(0, t) = 0, \quad y_2(1, t) = 0.55.$$

- $\gamma \gg 1$ for chemical stiffness; $\epsilon \ll 1$ for small diffusion
- Maas-Pope ILDM:

$$y_2 = \frac{y_1}{1 + y_1} + \frac{2y_1^2}{\gamma(\gamma - 1)(1 + y_1)^3}.$$

- *Purely* reactive system has equilibrium *point* in phase space at $y_1 = 0, y_2 = 0$ at $t \rightarrow \infty$.
- System with convection-diffusion approaches steady state *manifold*, not ILDM, as $t \rightarrow \infty$ given by solution of ODEs:

$$0 = -y_1 + \epsilon \frac{d^2 y_1}{dx^2}; \quad y_1(0) = 0; \quad y_1(1) = 1,$$

$$0 = -\gamma y_2 + \frac{(\gamma - 1)y_1 + \gamma y_1^2}{(1 + y_1)^2} + \epsilon \frac{d^2 y_2}{dx^2}; \quad y_2(0) = 0; \quad y_2(1) = 0.55.$$

Center manifold-motivated correction for small convection-diffusion

- Assume convection-diffusion acts as a small perturbation
- Define fast (w_f) and slow (w_s) variables based on analytic Jacobian of chemical source term:

$$\begin{pmatrix} w_s(x, t) \\ w_f(x, t) \end{pmatrix} = \begin{pmatrix} 1 & 0 \\ -\alpha(y_{10}(x)) & 1 \end{pmatrix} \cdot \begin{pmatrix} y_1(x, t) - y_{10}(x) \\ y_2(x, t) - h(y_{10}(x)) \end{pmatrix}$$

- $y_{10}(x)$ is solution for $y_1(x)$ at previous time step,
- $h(y_{10}(x))$ is the ILDM.
- Project original PDEs onto the slow and fast basis near ILDM to

get

$$\frac{\partial w_s}{\partial t} = -y_{10}(x) - w_s + \underbrace{\epsilon \left(\frac{d^2 y_{10}}{dx^2} + \frac{\partial^2 w_s}{\partial x^2} \right)}_{\text{convection-diffusion correction}} + H.O.T.,$$

$$\begin{aligned} \frac{\partial w_f}{\partial t} = & \underbrace{\text{Maas-Pope ILDM term}}_{=0} - \gamma w_f + \\ & + \underbrace{\epsilon \left(g_1(y_{10}(x)) + w_s g_2(y_{10}(x)) + \frac{\partial w_s}{\partial x} g_3(y_{10}(x)) + \frac{\partial^2 w_f}{\partial x^2} \right)}_{\text{convection-diffusion correction}} \\ & + H.O.T. \end{aligned}$$

Center manifold-motivated correction for small convection-diffusion

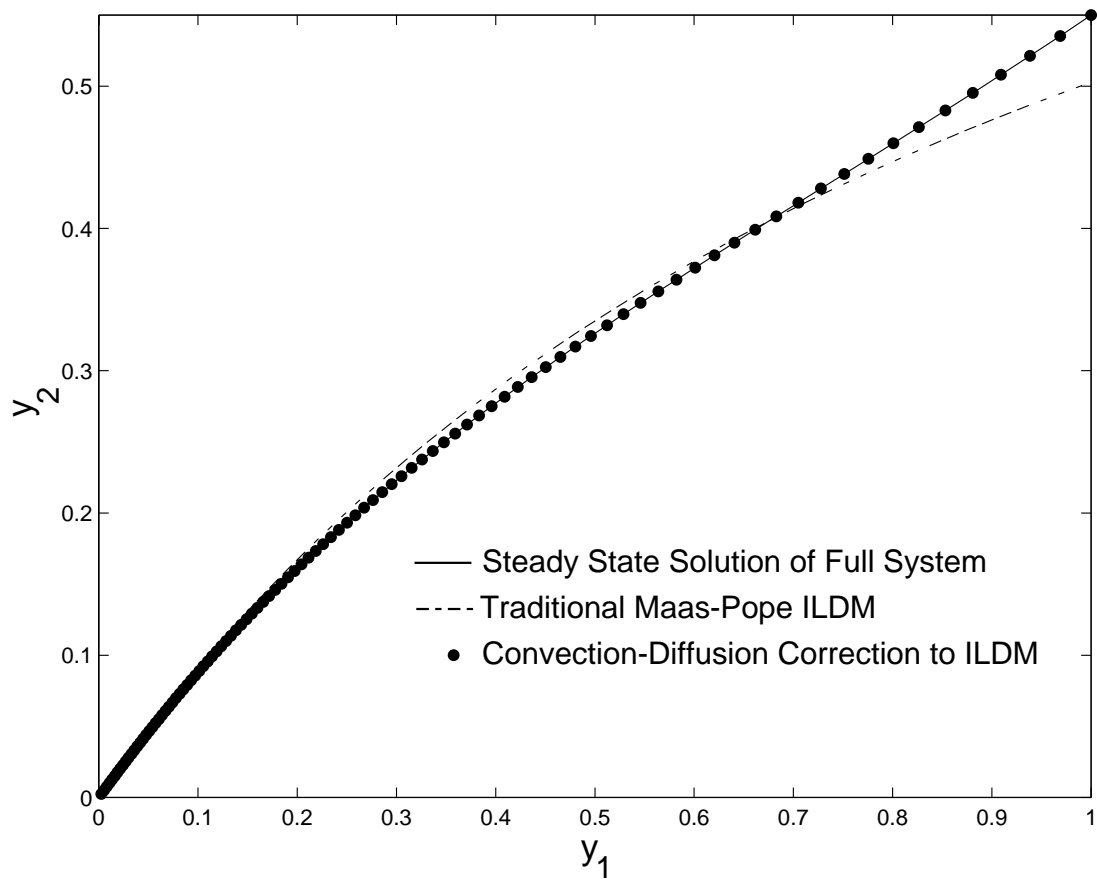
- equilibrate fast variables: $\frac{\partial w_f}{\partial t} = 0$, giving an elliptic equation

$$0 = -\gamma w_f + \underbrace{\epsilon \left(g_1(y_{10}(x)) + w_s g_2(y_{10}(x)) + \frac{\partial w_s}{\partial x} g_3(y_{10}(x)) + \frac{\partial^2 w_f}{\partial x^2} \right)}_{\text{convection-diffusion correction}} + H.O.T.$$

- Use method of lines, combined with simultaneous solution of elliptic equation, to advance slow variables using large time step,
- Analogous to solving elliptic equation for pressure when time advancing incompressible Navier-Stokes equations.

Center Manifold-Motivated Correction for Convection-Diffusion

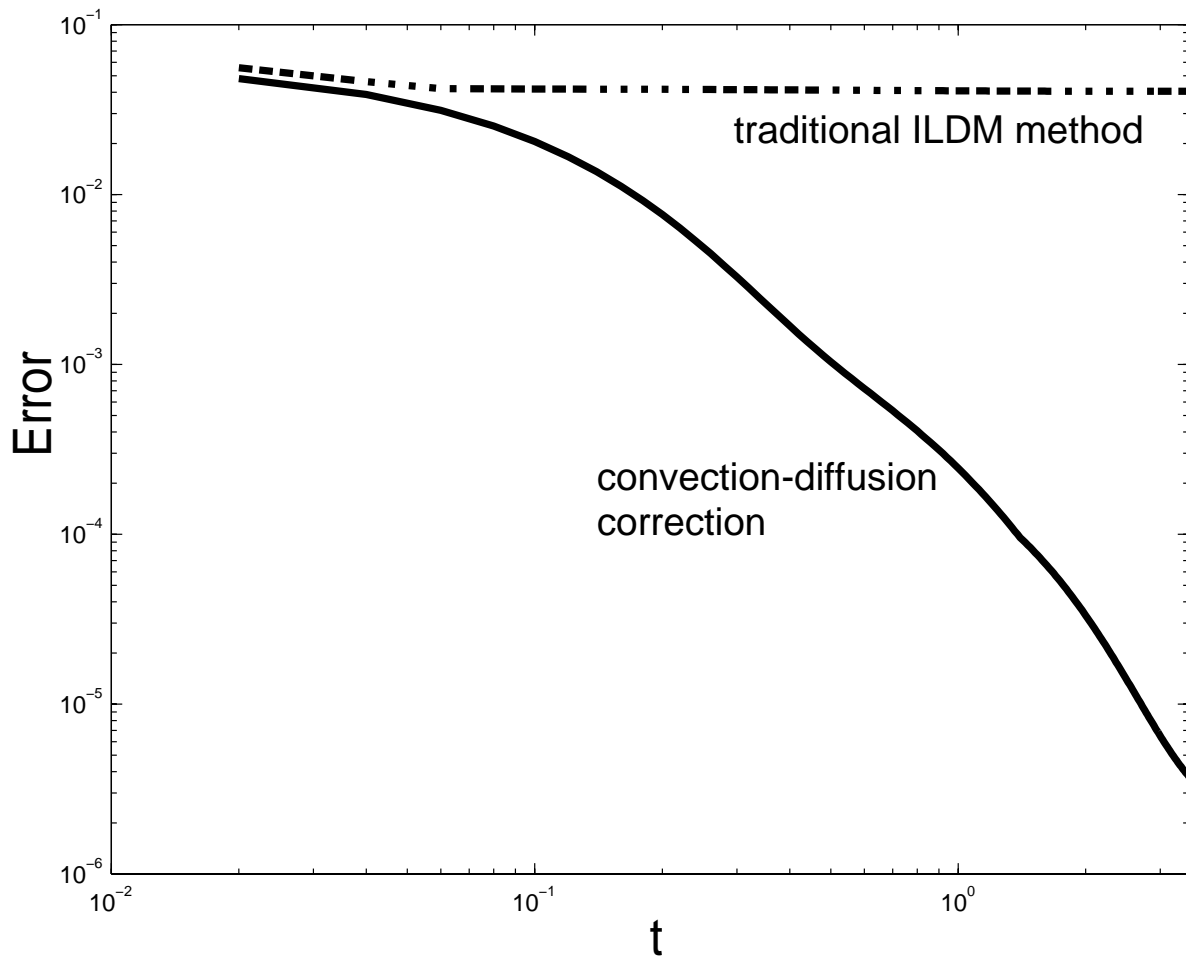
- Long time solution does not approach Maas and Pope ILDM,
- Convection-diffusion correction gives more accurate predictions



- $H_2/O_2/Ar$ ILDM results accurate because restricted to near equilibrium regions
- arbitrary use of ILDM can give inaccurate results

Center Manifold-Motivated Correction for Convection-Diffusion

- The corrected method gives more accurate predictions of intermediate and long times.



Conclusions

- The WAMR method gives dramatic spatial resolution in viscous one-dimensional H_2/O_2 detonations with detailed kinetics; viscous shocks, entropy layers, and induction zones are resolved,
- ILDM method, coupled appropriate use of full integration, with operator splitting accurately recovers most results of full chemistry with decrease (factor of three for our case) in computational time,
- Work needed to better account for projection of initial conditions onto ILDM,
- Center manifold-motivated correction for convection and diffusion a promising improvement to ILDM method in some (not all) problems.